Compound Semiconductors Materials Science

> MSE 5460 / ECE 5570 Spring, 2016

#### About the course



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#### MSE 5460 Compound Semiconductors Materials Science

#### Course information provided by the Courses of Study 2015-2016.

Course provides comprehensive understanding of compound semiconductors, stressing how differences with Si enable unique applications in electronics, photonics, and thermoelectrics. Materials include group IV (SiGe, SiC), III-V (GaAs, InP, GaN), II-VI (ZnO, CdTe), and layered semiconductors (Bi2Se3, BN, MoS2), and will touch on organic and dilute magnetic semiconductors. Methods of crystal growth, doping, strain control, and fabrication will be discussed. Electronic bandstructure will be used to develop band diagrams of complex heterostructures and nanostructures including quantum wells, wires, and dots. Examples of these materials in electronic, photonic, thermoelectric, and solar cell applications will be used to motivate the materials science.

When Offered: Fall or spring.

Prerequisites/Corequisites: Prerequisite: MSE 2620 or equivalent.

✓ Collapse Enrollment Information



#### Spring 2016 🗸

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# The first transistor





The Nobel Prize in Physics 1956 William B. Shockley, John Bardeen, Walter H. Brattain



#### The Nobel Prize in Physics 1956







William Bradford Shockley Prize share: 1/3

John Bardeen Prize share: 1/3

Walter Houser Brattain Prize share: 1/3

The Nobel Prize in Physics 1956 was awarded jointly to William Bradford Shockley, John Bardeen and Walter Houser Brattain "for their researches on semiconductors and their discovery of the transistor effect".

Photos: Copyright © The Nobel Foundation



#### First Demonstration of Amplification at 1 THz Using 25-nm InP High Electron Mobility Transistor Process

Xiaobing Mei, Wayne Yoshida, Mike Lange, Jane Lee, Joe Zhou, Po-Hsin Liu, Kevin Leong, Member, IEEE, Alex Zamora, Jose Padilla, Stephen Sarkozy, Richard Lai, Fellow, IEEE, and William R. Deal, Senior Member, IEEE

SUMMARY OF InP I

Gate length

Year Introduced

n(x)Ga(1-x)As channe

Source-Drain Spacing (µm)

indium composition

Abstract-We report the first ever terahertz monolithic integrated circuit amplifier based on 25-nm InP high electron mobility transistor (HEMT) process demonstrating amplification at 1 THz (1000 GHz) with 9-dB measured gain at 1 THz. This milestone was achieved with a 25-nm InP HEMT transistor, which exhibits 3.5-dB maximum available gain at 1 and 1.5 THz projected f<sub>MAX</sub>.

Index Terms-High electron mobility transistor (HEMT), high frequency, terahertz monolithic integrated circuit (TMIC), 25 nm gate, indium phosphide (InP).

#### I. INTRODUCTION

THE last decade has shown marked improvement in transistor speed capabilities. A variety of MMIC compatible processes, including HEMT and HBT, with a cut off frequency fT in excess of 600 GHz and a maximum frequency of oscillation f<sub>MAX</sub> approaching or exceeding 1.0 THz have been reported [1]-[5]. This improvement in capabilities has led to a rapid increase in integrated circuit capabilities, with amplifiers being demonstrated at frequencies in excess of 600 GHz [6]-[10].

					lds	400	
TA EMT Pr	BLE I ocesses	DEVELOF	PED AT N	GAS		200	
100nm	70 nm	35 nm	30 nm	25 nm (this work)			
1998	2003	2007	2010	2013		0	
60%	75%	100%	100%	100%			0.0
2	2	1.5	1.0	0.5			
0.12	0.1	0.04	0.04	0.04			

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of the model extraction. Further milestone achievements in amplification at higher frequencies with subsequent generation of transistors and designs at 480 GHz [11], 670 GHz [8] and 850 GHz [9] have improved the validity and accuracy







Fig. 4. Measured MAG/MSG and the associated  $h_{21}$  of a 2-finger 10  $\mu$ m device biased at  $V_d = 1.2$  V and  $I_d = 450$  mA/mm. The dashed lines represent data extrapolations with -20dB/decade and -10dB/decade slopes for the estimation of fT and fMAX.



Fig. 5. Microphotograph of 1.0 THz TMIC amplifier.



Fig. 6. On wafer measurement results showing 9 dB on-wafer gair at 1.0 THz.













**The Nobel Prize in Physics 1973** Leo Esaki, Ivar Giaever, Brian D. Josephson

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#### Leo Esaki - Facts



Leo Esaki

**Born:** 12 March 1925, Osaka, Japan

Affiliation at the time of the award: IBM Thomas J. Watson Research Center, Yorktown Heights, NY, USA

Prize motivation: "for their experimental discoveries regarding tunneling phenomena in semiconductors and superconductors, respectively"

Field: condensed matter physics, semiconductors

Prize share: 1/4

Science of electrons



VOLUME 45, NUMBER 6

PHYSICAL REVIEW LETTERS

11 AUGUST 1980

#### New Method for High-Accuracy Determination of the Fine-Structure Constant Based on Quantized Hall Resistance

K. v. Klitzing

Physikalisches Institut der Universität Würzburg, D-8700 Würzburg, Federal Republic of Germany, and Hochfeld-Magnetlabor des Max-Planck-Instituts für Festkörperforschung, F-38042 Grenoble, France

and

G. Dorda Forschungslaboratorien der Siemens AG, D-8000 München, Federal Republic of Germany

#### and

M. Pepper Cavendish Laboratory, Cambridge CB3 0HE, United Kingdom (Received 30 May 1980)

Measurements of the Hall voltage of a two-dimensional electron gas, realized with a silicon metal-oxide-semiconductor field-effect transistor, show that the Hall resistance at particular, experimentally well-defined surface carrier concentrations has fixed values which depend only on the fine-structure constant and speed of light, and is insensitive to the geometry of the device. Preliminary data are reported.

PACS numbers: 73.25.+i, 06.20.Jr, 72.20.My, 73.40.Qv

In this paper we report a new, potentially highaccuracy method for determining the fine-structure constant,  $\alpha$ . The new approach is based on the fact that the degenerate electron gas in the inversion layer of a MOSFET (metal-oxide-semiconductor field-effect transistor) is fully quantized when the transistor is operated at helium temperatures and in a strong magnetic field of order 15 T.<sup>1</sup> The inset in Fig. 1 shows a schematic diagram of a typical MOSFET device used in this work. The electric field perpendicular to the surface (gate field) produces subbands for the motion normal to the semiconductor-oxide interface, and the magnetic field produces Landau quantization of motion parallel to the interface. The density of states D(E) consists of broadened  $\delta$  functions<sup>2</sup>; minimal overlap is achieved if the magnetic field is sufficiently high. The number of states,  $N_{\rm L}$ , within each Landau level is given by

$$N_{\rm L} = eB/h, \qquad (1)$$

where we exclude the spin and valley degeneracies. If the density of states at the Fermi energy,  $N(E_{\rm F})$ , is zero, an inversion layer carrier cannot be scattered, and the center of the cyclo-





The Nobel Prize in Physics 1985 Klaus von Klitzing

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# The Nobel Prize in Physics 1985



Klaus von Klitzing Prize share: 1/1

The Nobel Prize in Physics 1985 was awarded to Klaus von Klitzing *"for the discovery of the quantized Hall effect"*.

Photos: Copyright © The Nobel Foundation

#### Science of electrons

#### Debdeep Jena (djena@cornell.edu), Cornell University

<u>38/xx</u>



 Low divergence circular beam (simplifies coupling to optical fibers)

"High performance laser to the cost of an LED"

Heterostructure containing about 200 epitaxial layers of different composition, thickness, and doping Layer thicknesses in the range 60-900 Å

Requirements in layer thickness precision =  $\pm 0.5$  %



The Nobel Prize in Physics 2000 Zhores I. Alferov, Herbert Kroemer, Jack S. Kilby

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# The Nobel Prize in Physics 2000







Zhores I. Alferov Prize share: 1/4

Herbert Kroemer Prize share: 1/4 Jack S. Kilby Prize share: 1/2

The Nobel Prize in Physics 2000 was awarded "for basic work on information and communication technology" with one half jointly to Zhores I. Alferov and Herbert Kroemer "for developing semiconductor heterostructures used in high-speed- and optoelectronics" and the other half to Jack S. Kilby "for his part in the invention of the integrated circuit".

Photos: Copyright © The Nobel Foundation

#### Science of electrons

#### Debdeep Jena (djena@cornell.edu), Cornell University

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**The Nobel Prize in Physics 2014** Isamu Akasaki, Hiroshi Amano, Shuji Nakamura

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# The Nobel Prize in Physics 2014



Photo: A. Mahmoud Isamu Akasaki Prize share: 1/3



Photo: A. Mahmoud Hiroshi Amano Prize share: 1/3



Photo: A. Mahmoud Shuji Nakamura Prize share: 1/3

The Nobel Prize in Physics 2014 was awarded jointly to Isamu Akasaki, Hiroshi Amano and Shuji Nakamura *"for the invention of efficient blue light-emitting diodes which has enabled bright and energy-saving white light sources"*.

#### Science of light

#### Debdeep Jena (djena@cornell.edu), Cornell University

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# Kroemer's 'theorem' on New Technologies







#### Compounds semiconductors: Science

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Science and Future

### Optical absorption and emission by atoms



Absorption and emission spectra of atoms was the first hint that electron energies are <u>quantized</u>.
Lead to the development of quantum mechanics, and finds a wide range of applications today.



# **Optical absorption in Compound Semiconductors**





- Absorption spectrum of typical compound semiconductors
- Can be completely explained using bandstructure+time-dep. Perturbation theory.



# **Optical absorption in Compound Semiconductors**





Absorption power is Quantified by "Absorption Coefficient"
A <u>negative</u> absorption coefficient is equivalent to <u>optical gain</u>!

• Adapted from: Wolfe/Holonyak/Stillman Physical Properties of Semiconductors



### Atoms to Crystals



### The elements

# **Periodic Table of Elements**

	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
1	1 H Hydrogen 1.00794	<sup>1</sup> Atomic # Symbol Name Atomic Mass	С	Solid				Metals			Nonme	tals						2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	2 K
2	3 Li Lithium 6.941	<sup>2</sup> 1 4 <sup>2</sup> 2 <b>Be</b> Beryllium 9.012182	Hợ H	Liquid Gas		Alkali me	Alkaline earth me	Lanthanoi	I ransition metals	Poor me	Other nonmeta	Noble ga	5 3 B Boron 10.811	6 24 C Carbon 12.0107	7 25 N Nitrogen 14.0067	8 26 O Oxygen 15.9994	9 27 F Fluorine 18.9984032	10 20 Neon 20.1797	2 K 8 L
3	11 Na Sodium 22.98976928	<sup>2</sup> 12 <b>Mg</b> Magnesium 24.3050	R	f Unknow	'n	tals	t <mark>als</mark>	Actinoids	د	tals	<u>v</u>	ses	13 28 Al Aluminium 26.9815386	14 28 Silicon 28.0855	15 <b>P</b> Phosphorus 30.973762	16 28 Sulfur 32.085	17 28 Cl Chlorine 35.453	18 Ar Argon 39.948	28 L 88 M
4	19 K Potassium 39.0983	<sup>2</sup> 20 28 Ca <sup>2</sup> <sup>8</sup> <sup>2</sup> <sup>2</sup> <sup>2</sup> <sup>2</sup> <sup>2</sup> <sup>2</sup> <sup>2</sup> <sup>2</sup> <sup>2</sup> <sup>2</sup>	21 28 29 2 20 20 20 20 20 20 20 20 20 20 20 20 2	22 <b>Ti</b> <sup>12</sup> <sup>13</sup> <sup>14</sup> <sup>15</sup> <sup>15</sup> <sup>15</sup> <sup>15</sup> <sup>15</sup>	23 <sup>2</sup> V <sup>11</sup> Vanadium 50.9415	24 28 Cr 13 Chromium 51.9981	25 <b>Mn</b> Manganese 54.938045	<sup>2</sup> <sup>2</sup> <sup>3</sup> <sup>1</sup> <sup>1</sup> <sup>1</sup> <sup>1</sup> <sup>1</sup> <sup>1</sup> <sup>1</sup>	<sup>2</sup> <sup>2</sup> <sup>4</sup> <sup>2</sup> <sup>2</sup> <sup>2</sup> <sup>2</sup> <sup>2</sup> <sup>2</sup> <sup>8</sup> <sup>15</sup> <sup>2</sup> <sup>2</sup> <sup>2</sup> <sup>8</sup> <sup>15</sup> <sup>2</sup> <sup>2</sup> <sup>8</sup> <sup>15</sup> <sup>2</sup>	28 <b>Ni</b> <sup>Nickel</sup> 58.6934	29 <b>Cu</b> Copper 63.546	30 <b>Zn</b> 21nc 85.38	31 28 Galium 69.723	32 28 Ge 4 Germanium 72.84	33 2 Ass <sup>18</sup> Arsenic 74.92160	34 28 Selenium 78.98	35 8 Br <sup>18</sup> Bromine 79.904	36 Kr Krypton 83.798	2000 8
5	37 <b>Rb</b> Rubidium 85.4678	38 28 5 5 5 5 5 5 5 7.62	39 2 <b>Y</b> 9 Yttrium 88.90585	40 <b>Zr</b> <sup>2</sup> irconium 91.224	41 28 Nb 18 Niobium 92.90638	42 2 8 Mo Molybdenum 95.96	43 <b>Tc</b> Technetium (97.9072)	44 <b>Ru</b> 101.07	<sup>2</sup> <sup>8</sup> <sup>8</sup> <sup>8</sup> <sup>8</sup> <sup>8</sup> <b>Rh</b> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup>	46 Pd Palladium 108.42	47 <b>Ag</b> Silver 107.8682	48 <b>Cd</b> 112.411	49 28 <b>In</b> 18 Indium 114.818	50 28 <b>Sn</b> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> 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6	55 <b>Cs</b> <sup>Caesium</sup> 132.9054519	<sup>2</sup> 56 <b>Ba</b> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup>	57–71	72 3 <b>Hf</b> 3 Hafnium 178.49	73 28 <b>Ta</b> 18 12 11 Tantalum 2 180.94788	74 28 13 12 12 12 12 12 12 12 12 12 12	75 <b>Re</b> Rhenium 188.207	<sup>2</sup> <sup>2</sup> <sup>2</sup> <sup>2</sup> <sup>2</sup> <sup>2</sup> <sup>3</sup> <sup>2</sup> <sup>3</sup> <sup>2</sup> <sup>3</sup> <sup>3</sup> <sup>3</sup> <sup>3</sup> <sup>1</sup> <sup>1</sup> <sup>1</sup> <sup>1</sup> <sup>1</sup> <sup>1</sup> <sup>1</sup> <sup>1</sup> <sup>1</sup> <sup>1</sup>	<sup>2</sup> <sup>8</sup> <sup>4</sup> <sup>2</sup> <sup>1</sup> <sup>16</sup> <sup>17</sup> <sup>18</sup> <sup>12</sup> <sup>15</sup> <sup>15</sup> <sup>15</sup> <sup>15</sup> <sup>15</sup> <sup>15</sup> <sup>15</sup> <sup>16</sup>	78 Pt Platinum 195.084	79 Au Gold 196.966569	80 80 <b>Hg</b> 1 Mercury 200.59	81 28 <b>TI</b> 18 Thallium 204.3833	82 2 <b>Pb</b> 32 Lead 4 207.2	83 28 Bi 18 Bismuth 208.98040	84 28 <b>Polonium</b> (208.9824)	85 28 At 18 Astatine 7 (209.9871)	86 <b>Rn</b> Radon (222.0178)	4882888
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		com		89 Ac	90 2 <b>Th</b> 18 10 10	91 28 Pa 322	92 U	<sup>2</sup> 93 <b>Np</b> <sup>1</sup> / <sub>2</sub>	<sup>2</sup> 94 <sup>2</sup> Pu <sup>32</sup> 22 Pu <sup>32</sup>	95 Am	96 Cm	97 Bk	98 <sup>2</sup> Cf <sup>10</sup>	99 28 Es 322	100 <sup>2</sup> <b>Fm</b> <sup>182</sup> <sub>30</sub>	101 <sup>2</sup> Md <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>32</sup>	102 <sup>2</sup> <b>No</b> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>18</sup> <sup>32</sup> <sup>32</sup>	103 Lr	2 8 8 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2

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2 232.03806 231.03588 238.02891

(237)

(244)

2 (243)

2 (247)

2 (247)

(257)

(252)

(259)



2

### Semiconductor Orbital Structures

#### Group IV

 $\mathbf{C}:[1s^2](2s^22p^2) \leftarrow 4$ 

$$Si:[1s^2][2s^2][2p^6](3s^23p^2) \leftarrow 4$$

 $\mathrm{Ge:}[1s^2][2s^2][2p^6][3s^2][3p^6][3d^{10}](4s^24p^2) \gets 4$ 

#### Group III

 $\mathbf{B}{:}[1s^2](2s^22p^1) \leftarrow \mathbf{3}$ 

Al:  $[1s^2][2s^2][2p^6](3s^23p^1) \leftarrow 3$ 

 $\mathbf{N}:[1s^2](2s^22p^3) \leftarrow 5$ 

P:  $[1s^2][2s^2][2p^6](3s^23p^3) \leftarrow 5$ 

 $\text{As:}[1s^2][2s^2][2p^6][3s^2][3p^6][3d^{10}](4s^24p^3) \leftarrow 5$ 











Find the eigenvalues of the Hamiltonian to get the bandstructure

$$\mathcal{E}_{\pm}(k_x, k_y) = A \pm \sqrt{B^2 + t^2} \left[ 1 + 4\cos(3k_x a/2)\cos(\sqrt{3}k_y a/2) + 4\cos^2(\sqrt{3}k_y a/2) \right]$$

$$A = (\epsilon_B + \epsilon_N)/2$$
  $B = (\epsilon_B - \epsilon_N)/2$ 







Find the eigenvalues of the Hamiltonian to get the bandstructure

$$\begin{split} \mathcal{E}_{\pm}(k_x,k_y) &= A \pm \sqrt{B^2 + t^2 \left[ 1 + 4\cos(3k_x a/2)\cos(\sqrt{3}k_y a/2) + 4\cos^2(\sqrt{3}k_y a/2) \right] } \\ \hline A &= (\epsilon_B + \epsilon_N)/2 \quad B = (\epsilon_B - \epsilon_N)/2 \\ \hline E_g &= \mathcal{E}_+ - \mathcal{E}_- = 2B = 5.8 \text{ eV} \\ \hline e_B &\sim +2.9 \text{ eV}, \epsilon_N &\sim -2.9 \text{ eV} \\ E_g &= 5.8 \text{ eV}, t = 2.92 \text{ eV}, a \sim 0.15 \text{ nm} \\ \hline Find the conduction and valence band structures and effective masses} \\ \hline \mathcal{E}_c(k) &= E_g + \frac{\hbar^2 k^2}{2m^\star}, \quad (\text{conduction band}) \\ \hline \mathcal{E}_v(k) &= -\frac{\hbar^2 k^2}{2m^\star}, \quad (\text{valence band}) \quad \hline h\omega_0 \\ \hline m^\star &= \frac{2\hbar^2 E_g}{9a^2t^2} \sim 0.6m_0 \\ \hline \end{split}$$

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### A Simple Example: 2D Graphene Bonds and Bands





## A Simple Example: 2D Graphene Bonds and Bands



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## A Simple Example: 2D Graphene Bonds and Bands



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# Typical 3D semiconductor crystal structures

Due to the covalent bond structure (the bonding orbitals are s-p hybrids), most common semiconductors are found in three distinct structures:

- i) The diamond lattice type: e.g. Si, Ge
- ii) The zinkblende type: e.g. GaAs, InP
- iii) The wurtzite type (hexagonal): e.g. GaN, ZnO





Fig. 1.2 Diamond lattice structure.





Fig. 1.3 (s) Zincblende lattice structure and (b )Wurtzite lattice structure. ( $\bigcirc$ ) A atom, ( $\bigcirc$ ) B atom.

(Denninger, Oda)

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### Semiconductor Crystal Structures

Which electrons have to be included in band structure calculations?

In principle all, but the inner core electrons are strongly bound and their spatial extension is so small, that no significant overlap of the wavefunctions occur.

In silicon (Si), the 1s, 2s and 2p electrons are inner core electrons.





### Semiconductor Crystal Structures



8 electrons in 4 bands

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(Denninger)

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## Nearly free electron bandstructure, k-space points



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(Rockett)

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### Orbital overlaps and bonds in semiconductors





Fig. 2.17a,b. Overlap of two s orbitals to form bonding and antibonding  $\sigma$  orbitals



**Fig. 2.18a,b.** Overlap of two  $p_x$  orbitals along the x axis to form bonding and antibonding  $\sigma$  orbitals



**Fig. 2.19a,b.** Overlap of two  $p_y$  orbitals to form bonding and antibonding  $\pi$  orbitals

(Cardona/Yu)



### Orbital overlaps and bonds in semiconductors





 $\langle s | H | p_{y} \rangle$ 

 $V_{sp\sigma} \cos\Theta$ 

=

+

 $0\sin\Theta$ 



(Cardona/Yu)

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### Tight-Binding Bandstructure Matrix

$$\mathbf{a}_1 = \frac{a}{4}(1,1,1), \ \mathbf{a}_2 = \frac{a}{4}(-1,-1,1), \ \mathbf{a}_3 = \frac{a}{4}(-1,1,-1), \ \mathbf{a}_4 = \frac{a}{4}(1,-1,-1)$$



	$ s^A angle$	$ p_x^A angle$	$ p_y^A angle$	$ p_z^A angle$	$ s^B angle$	$ p_x^B angle$	$ p_y^B angle$	$ p^B_z angle$
$\langle s^A  $	$\int E_s^A$	0	0	0	$-V_0g_0({f k})$	$V_1g_1({f k})$	$V_1g_2({f k})$	$V_1g_3(\mathbf{k})$
$\langle p_x^A  $	0	$E_p^A$	0	0	$-V_1g_1({f k})$	$V_2 g_0({f k})$	$V_3g_3({f k})$	$V_3g_2({f k})$
$\langle p_y^A  $	0	0	$E_p^A$	0	$-V_1g_2({f k})$	$V_3g_3({f k})$	$V_2 g_0({f k})$	$V_3g_1({f k})$
$\langle p_{m{z}}^{A} $	0	0	0	$E_p^A$	$-V_1g_3({f k})$	$V_3g_2({f k})$	$V_3g_1({f k})$	$V_2 g_0({f k})$
$\langle s^B  $	c.c.	c.c.	c.c.	c.c.	$E_s^B$	0	0	0
$\langle p_x^B  $	c.c.	c.c.	c.c.	c.c.	0	$E_p^B$	0	0
$\langle p_y^B  $	c.c.	c.c.	c.c.	c.c.	0	0	$E_p^B$	0
$\langle p_{z}^{B} $	$\backslash$ c.c.	c.c.	c.c.	c.c.	0	0	0	$E_p^B$





Origin of s and p 'contents' in semiconductor bandstructure





ECE 407 – Spring 2009 – Farhan Rana – Cornell University











#### 5. ENGINEERING ELECTRONIC STRUCTURE

5.1 Linking atomic orbitals to bands	
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#### Rockett: MSS



*Figure 5.2:* Shows the symmetry of the hybridized  $sp^3$  and  $sp^2$  molecular orbitals. The  $sp^2$  orbitals lie in a plane perpendicular to the  $p_z$  orbitals and are equal lengths. The  $sp^3$  orbitals are all equivalent to each other and stretch to corners of a tetrahedron.

Origin of s and p 'contents' in semiconductor bandstructure

**195** ...196 ...197 ...201 ...206 ...215 ...223 ...226 ...226 ...226 ...226 ...226 ...228 ...228 ...229 ...229 ...220 ...226 ...226 ...226 ...226 ...228 ...228 ...229

> Figure 5.1: Shows the shape of the s and  $p_x$  orbitals and the equations that describe them.  $a_0$  is the atomic orbital size, r is the radius from the nucleus, and  $\theta$  is the angle in the x,y plane.

Wave function amplitudes (probability of finding an electron)



*Figure 5.3:* A schematic diagram illustrating the basis of cohesion in solids resulting from symmetric and antisymmetric combinations of atomic orbitals. The center of electron charge lies between the positive ions for a symmetric bonding orbital and outside of the positive ions for an antisymmetric orbital combination.



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*Figure 5.4:* A schematic diagram of the evolution of bonding of Si atoms. The filled 3s and partially filled 3p atomic orbitals of two atoms combine to form half-filled  $sp^3$  hybrid molecular orbitals. These combine to form bonding and antibonding orbitals. As more atoms collect atoms collect to create a bulk solid, bands form.

Silicon



*Figure 5.5:* Shows the evolution of atomic orbital energies to form bonds and ultimately bands. The geometries of the atomic and hybrid orbitals are shown schematically as insets.

GaAs






*Figure 5.6:* A schematic diagram showing the interactions of selected atomic orbitals and the geometry of these orbitals with respect to the crystal lattice in a zincblende or diamond structure material.

Semicon-	Semiconductor	Lattice	Energy Gap, eV	E <sub>c</sub>	E <sub>v</sub>
ductor Class		Parameter	(at 20°C)	(eV)	(eV)
Cubic		nm			
IV	C (diamond)	0.35597	5.5		
IV	Si	0.54307	1.12 (indirect)	4.05	5.17
IV	Ge	0.56754	0.67 (indirect)	4.0	4.67
IV	a-Sn	0.64912	0.08		
III-V	GaP	0.54505	2.26 (indirect)	3.8	6.1
III-V	GaAs	0.56532	1.42	4.07	5.49
III-V	GaSb	0.609593	0.726	4.06	4.79
III-V	InP	0.58687	1.344	4.38	5.72
III-V	InAs	0.60583	0.354	4.9	5.25
III-V	InSb	0.6479	0.17	4.59	4.76
II-VI	ZnSe (cubic)	0.567	2.58	4.1	6.7
I-VII	CuBr	5.69	2.94	4.35	7.29
I-III-VI <sub>2</sub>	CuInSe <sub>2</sub>	0.578	0.98	4.0	5.0
II-IV-V <sub>2</sub>	ZnGeAs <sub>2</sub>	0.567	0.85		
Hexagonal					
III-V	AlN	0.3111 (a)	5.9	0.6	6.5
		0.4978 (c)			
III-V	GaN	0.3190 (a)	3.45	4.0	7.4
		0.5189 (c)			
III-V	InN	0.3533 (a)	0.7 (note values		
		0.5693 (c)	vary greatly)		
II-VI	ZnS	0.3814 (a)	3.911		
		0.6258 (c)			
II-VI	CdSe	0.4299 (a)	1.751 eV		
		0.7010 (c)			
$E_c$ (the electron affinity) and $E_v$ (electron affinity + energy gap) measured with respect to					
the vacuum level. Lattice parameters in nm.					

Chemical bonding, Lattice Constants, Band edges, and Band Gaps for various Semiconductors

(Rockett)









Fig. 2.25. A comparison between the band structure of Ge calculated by (a) the tight-binding method. (b) the empirical pseudopotential method, and (c) the nearly free electron model [Ref. 2.24, p. 79]



**Fig. 2.24.** The valence band structure and density of states (see Sect. 4.3.1 for definition) of Si calculated by the tight-binding method (*broken curves*) and by the empirical pseudopotential method (*solid lines*) [2.25]

**Bandstructures and Density of States** 



#### Electron clouds in semiconductors





*Electron wavefunction squared = probability density of finding electrons* 

(Rockett)



#### Measurement of Semiconductor Bandstructures



Angle-Resolved Photo-Emission Spectroscopy (ARPES)

(Rockett)





FIGURE 2.16. Band structure of four common semiconductors: silicon, germanium, gallium ansemide, and aluminium arsemide. The calculations do not include the spin-orbit coupling. [Results kindly supplied by Prof. G. P. Srivastava, University of Exeter.]



### Semiconductor dielectric constants & polarization



#### Semiconductor density of states



#### Semiconductor carrier statistics

$$\begin{aligned} & \text{First Revealed to the devices of the set of the$$

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#### Semiconductor carrier statistics





#### Semiconductor doping



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## Semiconductor doping

Graphical Solution for Locating the Fermi Level





## Semiconductor doping

#### Controlling resistivity of semiconductors by doping





Resistivity of • Metals ~  $10^{-6}$  Ohm-cm • SiO<sub>2</sub> (insulator) ~  $10^{14}$  Ohm-cm

#### Various methods of "Doping"

#### Traditional Bulk Semiconductors -

- Impurity doping (bulk, uniform)
- Impurity doping (graded, delta-doping) (Since '50s)

#### Impurity + Heterostructures -

 Modulation Doping (Since '80s)

#### Polar Semiconductor Heterostructures -

 Polarization-induced doping (Since late '90s)

#### Other "Dynamic" methods -

- Electrostatic gating
- Optical excitation
- Thermal excitation

These are the various methods for Creating mobile carriers in the conduction & valence bands



### Mobility and Transport





#### **Compound Semiconductor Families**



Fig. 7.6. Bandgap energy and lattice constant of various III-V semiconductors at room temperature (adopted from Tien, 1988).

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#### **Compound Semiconductor Families**



Fig. 7.6. Bandgap energy and lattice constant of various III-V semiconductors at room temperature (adopted from Tien, 1988).

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### Compound Semiconductor Heterostructures: Strain



FIG. 1. (Color online) AFM images showing  $2 \times 2 \ \mu m^2$  scans of the AlN surfaces after growth.

APPLIED PHYSICS LETTERS 90, 182112 (2007)



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# Compound semiconductors: Heterostructures

#### Semiconductor Heterostructures



Semiconductor bands line up with each other in 3 ways.

Based on the lineups, one can create

- Barriers
- Wells
- Quasi-Electric fields

For <u>both electrons and holes independently</u> with suitable material choices.

Example – Double Heterostructure LED/Laser

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Staggered

Broken

90/xx

Straddling

# Being the Quantum Mechanic



#### Being the Quantum Mechanic



The particle in a box problem



**FIGURE 4.2.** Graphical solution of equation (4.12) for a square well in GaAs with depth  $V_0 = 0.3 \text{ eV}$  and width a = 10 nm, giving  $\theta_0^2 = 13.2$ . There are three bound states.



**FIGURE 4.3.** Local density of states n(E, z = 0) in the middle of a square well in GaAs of width 10 nm and depth 0.3 eV. The result for free electrons, proportional to  $E^{-1/2}$ , is shown for comparison.

(Rockett/Davies)

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#### Being the Quantum Mechanic: Quantum well states



**FIGURE 1.3.** Optical absorption in a quantum well formed by a layer of GaAs surrounded by AlGaAs. (a) Potential well in conduction and valence band, showing two bound states in each; the energy gap of GaAs is really much larger than this diagram implies. (b) Transitions between states in the quantum well produce absorption lines between the band gaps of the GaAs well and AlGaAs barrier.

Quasi-2D systems: Quantum Wells



**FIGURE 1.4.** Photoluminescence as a function of wavelength for a sample with four quantum wells of different widths, whose conduction and valence bands are shown on the right. The barriers between the wells are much thicker than drawn. [Data kindly supplied by Prof. E. L. Hu, University of California at Santa Barbara.]

Photoluminescence spectra of quantum wells





## Being the Quantum Mechanic: Designer Quantum wells



Parabolic quantum wells: Harmonic Oscillator States!

(Davies)

95/xx

# Being the Quantum Mechanic: Multiple Quantum Wells



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(Rockett) 96/xx

### The Quantum Mechanic: Designer Quantum States



FIG. 3. (a) TEM characterization of a test sample grown by MOCVD. The growth sequence and the structure are shown in (b). The remarkable features are the sharpness of the very narrow GaAs layers (minimum  $\approx 25$  Å) appearing at the lower right-hand side corner, the interface roughness showing up at the uppermost interface of the AlAs layer, and the subsequent smoothing of this roughness by the multilayer growth (upper left-hand side corner) (after Leys *et al.*<sup>61</sup>).

(Weisbuch/Vinter)



#### Heterostructure superlattices

SUPERLATTICES



N WELLS N-DEGENERATE GROUND STATE TIGHT-BINDING APPROXIMATION

$$\Psi_{q}(z) = \frac{1}{N^{1/2}} \sum e^{iqnd} \chi_{loc}(z-nd)$$

$$E = E_{1} + S + 2Tcosqd$$

$$s = \int X_{loc}(z-d)V(z)X_{loc}(z-d)dz$$

$$T = \int X_{loc}(z-d)V(z)X_{loc}(z)dz$$

$$AE = \sum_{z=1}^{2N \text{ states using BVK}} E^{zN \text{ states using BVK}}$$



FIG. 15. Tight-binding model of superlattices.



FIG. 16. Tight-binding model of GaAs/GaAlAs superlattices: Variation of the fundamental state bandwidth [4 $t_i$  of Eq. (30)] in the tight-binding model as a function of barrier thickness for three different well thicknesses. X = 0.2;  $V_0 = 212$  meV (after Bastard<sup>125</sup>).

(Weisbuch/Vinter)

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#### Heterostructure superlattices



FIG. 15. Tight-binding model of superlattices.



FIG. 18c. Allowed energy bands  $E_1$ ,  $E_2$ ,  $E_3$ , and  $E_4$  (hatched) calculated as a function of well or barrier width ( $L_e = L_B = a$ ) in a superlattice with a barrier potential V = 0.4 V. Note the existence of forbidden gaps even above the barrier potential (Reprinted with permission from World Scientific Pub. Co., L. Esaki, "Recent Topics in Semiconductor Physics" (H. Kamimura and Y. Toyozawce, eds.), 1983.)



FIG. 17. Comparison of the DOS of a superlattice with that of a 2D system (---) and a 3D isotropic system. Note the broadening of the superlattice band with band index as the overlap of wave functions increases with energy E in the tight-binding description, increasing the transfer matrix element  $t_i$  (Reprinted with permission from World Scientific Pub. Co., L. Esaki, "Recent Topics in Semiconductor Physics" (H. Kamimura and Y. Toyozawa, eds.), 1983.)

#### (Weisbuch/Vinter)

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#### Lower dimensions have lower DOS!





#### **Overview of Semiconductor Devices**



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#### P-n diode electrostatics



# Breakdown voltage of pn diodes



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## Electrostatics and current flow in a p-n diode



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# Compound semiconductors: Heterostructures

#### Semiconductor Heterostructures



Semiconductor bands line up with each other in 3 ways.

Based on the lineups, one can create

- Barriers
- Wells
- Quasi-Electric fields

For <u>both electrons and holes independently</u> with suitable material choices.

Example – Double Heterostructure LED/Laser

-00000

Staggered

В

Type II

A

Broken

B

BROKEN GROUP

TYPE III

Light

Acceptors

Straddling

 $E_c^B$ 

 $E_g^B$ 

 $E_v^B$ 

Electrons

Donors

 $\oplus$   $\oplus$   $\oplus$ 

 $E_c^A$ 

 $E_{\nu}^{A}$ 

Type i



# Compound semiconductor heterostructure band offsets







 $\Delta E_V(A:B) + \Delta E_V(B:C) + \Delta E_V(C:A) = 0.$ 

Transitivity of band offsets

The Common Anion Rule: When the anion (the electron accepting atom such as As in GaAs and InAs) is in common across a semiconductor heterojunction, the change in the conduction band edge is greater than the change in the valence band edge across the semiconductor heterojunction. Mathematically,  $\Delta E_V < \Delta E_C$ .

**The Common Cation Rule**: When the cation (e.g. Ga in GaAs or GaSb) is in common across the junction, the valence band edge energies scale with the anion electronegativities. For example, the valence band edge of phosphide semiconductors will lie below those for arsenides which will lie below those of antimonides. Mathematically,  $E_V(CA_1) < E_V(CA_2) < E_V(CA_3)$ , where C designates a cation, and  $A_1$ ,  $A_2$ , and  $A_3$  designate three anions with decreasing electronegativities.





(Rockett)



### Quasi electric fields in semiconductor heterostructures

Semiconductor Heterostructures & Quasi-Electric Fields



in manun manun ... a drift field may also be generated through a variation of the energy gap itself, by making the base region from a nonstoichiometric mixed crystal of different semiconductors with different energy gaps (for Fe = O example, Ge-Si), with a composition that var-777777777777777777777777777777777777 ies continuously through the base. (Translated from Krömer, 1954) MILLIUM qFh Forces different for electrons and holes 11111 Rev. Mod. Phy., vol 73, p 783 (2001)

FIG. 1. Quasielectric fields: (a) A true electric field simply tilts the bands; (b) quasielectric fields, with no force on electrons, but a force on holes; (c) quasielectric fields forcing electrons and holes in the same direction. From Kroemer, 1957a.



Equal force on electrons and holes



#### Compound semiconductors: Heterostructures

# Nobel Lecture: Quasielectric fields and band offsets: teaching electrons new tricks\*

#### Herbert Kroemer

*Electrical and Computer Engineering Department, University of California, Santa Barbara, California 93106-9560* 

(Published 22 October 2001)

#### I. INTRODUCTION

Heterostructures, as I use the word here, may be defined as heterogeneous semiconductor structures built from two or more different semiconductors, in such a way that the transition region or interface between the different materials plays an essential role in any device action. Often, it may be said that the interface is the device.

The participating semiconductors all involve elements from the central portion of the periodic table of the elements (Table I). In the center is silicon, the backbone of modern electronics. Below Si is germanium. Although Ge is rarely used by itself, Ge-Si alloys with a composition-dependent position play an increasingly important role in today's heterostructure technology. In fact, historically this was the first heterostructure device system proposed, although it was also the system that took longest to bring to practical maturity, largely because of the 4% mismatch between the lattice constants of Si and Ge.

Silicon plays the same central role in electronic metallurov that steel plays in structural metallurov But just Similar to the III-V compounds, every element shown in column II may be used together with every element in column VI to create II-VI compounds, and again alloying is possible to create a continuous range of the latter.

#### II. BAND DIAGRAMS AND QUASIELECTRIC FORCES

Whenever I teach my semiconductor device physics course, one of the central messages I try to get across early is the importance of energy-band diagrams. I often put this in the form of "<u>Kroemer's Lemma of Proven</u> Ignorance":

If, in discussing a semiconductor problem, you cannot draw an *Energy-Band-Diagram*, this shows that you don't know what you are talking about,

with the corollary

If you can draw one, but don't, then your audience won't know what you are talking about.

Nowhere is this more true than in the discussion of heterostructures, and much of the understanding of the latter is based on one's ability to draw their band diagrams—and knowing what they mean.

• Rev. Mod. Phys., vol 73, pg. 783 (2001)

# Heterojunction Energy-Band Diagrams





How to draw the energy band diagram for a heterostructure

(Rockett)

Debdeep Jena (djena@cornell.edu), Cornell University

dopant


### High Electron Mobility Transistors (HEMTs, or HFETs)

# Conductive channel is a 2DEG in a heterostructure quantum well







#### Key Advantages of HFETs:

- High mobility (suppressed ionized impurity scattering)
- Superior low-temperature performance (no freezeout)
- Choices: superior channel materials, heterostructures
- High sheet charge densities & channel conductance

#### Key material Requirements for HFETs:

- Semi-insulating substrates
- Large band offsets for carrier quantum confinement
- High mobility in conductive channel
- Low resistance Ohmic contacts
- Low-leakage Gate (Schottky, MOS structure, or variants)

### HEMTs / HFETs

#### Semiconductor combinations for HFETs





Fig. 7.6. Bandgap energy and lattice constant of various III-V semiconductors at room temperature (adopted from Tien, 1988). Debdeep Jena (alena@cornell.eau), Cornell University



Fig. 12.12. Bandgap energy versus lattice constant of III–V nitride semiconductors at room temperature.



### HEMTs / HFETs

#### Semiconductor combinations for HFETs





Fig. 7.6. Bandgap energy and lattice constant of various III-V semiconductors at room temperature (adopted from Tien, 1988). Debdeep Jena (djena@cornell.edu), Cornell University



Fig. 12.12. Bandgap energy versus lattice constant of III–V nitride semiconductors at room temperature.

#### Major HFET families:

- GaAs-based (AlGaAs/GaAs) (strain-free, or pseudomorphic)
- InP based (InGaAs channels)
- 6.1 Angstrom/narrow gap channels (generally grown metamorphically on GaAs)

113/xx

• GaN-based (AIGaN/GaN and AIN/GaN) (typically pseudomorphic)

### HEMTs / HFETs

#### Semiconductor combinations for HFETs



Ultra-fine gate structure

#### HEMTs / HFETs: 2D Electron Gases

#### Charge Control Model

Charge Neutrality: 
$$N_d^+ = n_m + n_{par} + n_s$$

Gate Capacitance:  $C_G = \frac{\epsilon}{d + \Delta d} \longleftrightarrow V_{GS,int} = V_{GS} - (E_{F,ch} - E_{F,s})$ 'Quantum' Capacitance:  $C_Q = \epsilon/\Delta d \approx q^2 \times DOS$  $C_G = \frac{C_{G0}C_Q}{C_{C0} + C_O}$ 

$$\mathcal{C}_Q = \epsilon / \Delta d \approx q^2 \times DOS$$
  
 $\rightarrow \Delta d \approx rac{\epsilon \pi \hbar^2}{q^2 m^{\star}} \sim 8 \text{ nm (GaAs)}$   
 $\sim 2 \text{ nm (GaN, Si MOSFETs)}$ 

Effect of low Quantum Capacitance (low effective mass):

- More gate voltage needed for same carrier concentration
- Bad for gate modulation & transconductance
- This is a feature of 2DEGs (1D nanostructures can eliminate QC!)

#### Debdeep Jena (djena@cornell.edu), Cornell University







(d)

 $F_1 = \frac{-en_m}{2}$ 

### HEMTs / HFETs: Threshold Voltage



QW energy eigenvalues (Airy functions):  $E_i \simeq \left(\frac{\hbar^2}{2m^*}\right)^{1/3} \left[\frac{3}{2}eF_2\pi\left(i+\frac{3}{4}\right)\right]^{2/3}$ 





Voltage drop in QW due to quantum capacitance:

$$qV_{di}^{-} \approx E_F - E_0 = \frac{\pi\hbar^2}{m^{\star}} n_s$$



(c)

(d)

F(z)

 $F_2 = \frac{en_s}{s}$ 

 $F_1 = \frac{-en_m}{c}$ 

Ζ

Energy Conservation:

$$-V_G + \phi_b - V_1 + V_2 - \frac{\Delta E_c}{e} + V_{di}^- = 0$$

Gauss's Law:  

$$V_{1} = \frac{en_{m} (d - d_{s})}{\epsilon}$$

$$V_{2} = \frac{en_{s}d_{s}}{\epsilon}$$

$$q\Delta n_{s} = (\frac{\epsilon}{d + \Delta d})\Delta V_{G} = C_{G}\Delta V_{G} \overset{e_{F,s}}{\underset{-d}{\overset{-e_{V_{G}}}{\overset{-e_{V}}{\overset{-e_{V}}{\overset{-e_{V}}}{\overset{-e_{V}}{\overset{-e_{V}}}{\overset{-e_{V}}{\overset{-e_{V}}}{\overset{-e_{V}}{\overset{-e_{V}}}{\overset{-e_{V}}{\overset{-e_{V}}}{\overset{-e_{V}}{\overset{-e_{V}}}{\overset{-e_{V}}{\overset{-e_{V}}}{\overset{-e_{V}}{\overset{-e_{V}}}{\overset{-e_{V}}{\overset{-e_{V}}}{\overset{-e_{V}}{\overset{-e_{V}}}{\overset{-e_{V}}{\overset{-e_{V}}}{\overset{-e_{V}}{\overset{-e_{V}}}{\overset{-e_{V}}{\overset{-e_{V}}}}{\overset{-e_{V}}}{\overset{-e_{V}}}{\overset{-e_{V}}}{\overset{-e_{V}}}{\overset{-e_{V}}}{\overset{-e_{V}}}{\overset{-e_{V}}}}{\overset{-e_{V}}}{\overset{-e_{V}}}{\overset{-e_{V}}}{\overset{-e_{V}}}{\overset{-e_{V}}}}{\overset{-e_{V}}}{\overset{-e_{V}}}{\overset{-e_{V}}}{\overset{-e_{V}}}{\overset{-e_{V}}}{\overset{-e_{V}}}{\overset{-e_{V}}}{$$

Charge control Law:

$$n_s(V_G) = \frac{eN_d^+ d_\delta + \epsilon \left[V_G - (\phi_b - \Delta E_c/e)\right]}{eD}$$

Threshold Voltage (n<sub>s</sub>=0)  $\rightarrow V_p = -\frac{eN_d^+ d_{\delta}}{\epsilon} + (\phi_b - \Delta E_c/e)$ 

#### **Overview of Semiconductor Devices**



*Figure 3.17:* The four possible metal-semiconductor cases before and after contact. Two of the four produce ohmic contacts (resistors) while the others produce Schottky diodes.



(Rockett)



### **Compound Semiconductor Heterojunction Devices**





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### **Compound Semiconductor Heterojunction Devices**



(Rockett)

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### GaN family (Wurtzite III-Vs)



- III-V nitride semiconductors crystallize in the wurtzite structure
- All three semiconductors (InN, GaN, and AIN) are direct gap
- Bandgaps range from the IR thru visible to deep UV



### Spontaneous & Piezoelectric polarization in the Nitrides



- III-V Nitrides have an order of magnitude higher piezo-polarization than ZB III-Vs
- Piezoelectric polarization of Al(Ga)N is higher than GaN
- Alloys scale roughy linearly with composition (Vegard's law)















Perfect Tetrahedron: Dipoles cancel exactly

If 
$$\left[\frac{c}{a} = \sqrt{\frac{8}{3}}\right]$$
,  $\sum \mathbf{p}_i = 0!$ 









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### GaN HEMTs: Consequence of Polarization Fields

#### Consequences of Polarization



### Polar Heterostructures

#### Charge Control Model



Polarization sheet charge: 
$$Q_{\pi}(\text{net}) = Q_{\pi}(\text{AlGaN}) - Q_{\pi}(\text{GaN})$$

about abarrage ( (mat)

Energy conservation (ungated): 
$$\phi_s - V_1 - \frac{\Delta E_c}{e} + V_{di}^- = 0$$

Gauss's Law:

$$V_1 = \frac{\left[Q_\pi(\text{net}) - en_s\right] d_{AlGaN}}{\epsilon}$$





$$n_s = \frac{Q_{\pi}(\text{net}) \cdot d_{AlGaN} - \epsilon \, \phi_s - \Delta E_c/e)}{eD}$$

Charge density (gated):

$$n_s(V_G) = \frac{Q_{\pi}(\text{net}) \cdot d_{AlGaN} + \epsilon \left[V_G - \left(\phi_b - \Delta E_c/e\right)\right]}{eD}$$



GaN-based polar HFETs:

- Wide bandgap >> High breakdown voltages
- Wide bandgap >> Thermal stability

• 2DEG channel due to polarization (very high densities possible)

du), Cornell University



 $O(O \cdot \mathbf{N})$ 

#### Polarization charges at Heterojunctions





### Polarization charges @ heterojunctions: Consequences



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### Polarization Engineering in III-Nitride Heterostructures



Polarization creates mobile electron and hole gases: WITHOUT DOPING!



"Dipole Engineering"-Polarization dipoles can create artificial band-offsets in GaN heterostructures



**Figure 2** An ultrathin polarization dipole layer results in a quasi band-offset leading to staggered or broken gap alignments.

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#### Polarization and Band Parameters for III-Nitrides

#### Parameters for InN, GaN, AlN, and Their Ternary Compounds

Parameters	GaN	AIN	InN
a (Å) at $T = 300$ K	3.189	3.112	3.545
c (Å) at $T = 300 K$	5.185	4.982	5.703
$E_g$ (eV) at $T = 0$ K	3.510	6.25	0.69 <sup>b</sup>
at $T = 300 \text{K}$	3.44	6.16	0.64
$\alpha (\text{meV/K})$	0.909	1.799	0.41 <sup>b</sup>
β(K)	830	1462	454 <sup>b</sup>
$\Delta_{cr}$ (eV)	0.010	-0.169	0.040
$\Delta_{so} (eV)$	0.017	0.019	0.005
$m_e^{\parallel}/m_0$	0.20	0.32	0.07
$m_{e}^{\perp}/m_{0}$	0.20	0.30	0.07
$A_1$	-7.21	- 3.86	-8.21
A2	-0.44	-0.25	-0.68
A3	6.68	3.58	7.57
$A_4$	-3.46	-1.32	-5.23
A5	-3.40	-1.47	-5.11
A <sub>6</sub>	-4.90	-1.64	- 5.96
$A_7 (\text{eV} \text{\AA})$	0.0937	0	0
$a_1 (eV)$	-4.9	-3.4	-3.5
$a_2 (eV)$	-11.3	-11.8	- 3.5
$D_1$ (eV)	-3.7	-17.1	-3.7
$D_2$ (eV)	4.5	7.9	4.5
$D_3$ (eV)	8.2	8.8	8.2
$D_4 (eV)$	-4.1	-3.9	-4.1
$D_5 (eV)$	-4.0	-3.4	-4.0
$D_6 (eV)$	-5.5	-3.4	-5.5
C11 (GPa)	390	396	223
C12 (GPa)	145	137	115
C13 (GPa)	106	108	92

General Interpolation Formula for Ternary Compound Parameters P: $P(A_xB_{1-x}C) = xP(AC) + (1-x)P(BC)$ Except Band-Gap Energy: $P(A_xB_{1-x}C) = xP(AC) + (1-x)P(BC) - bx(1-x)$			
In <sub>x</sub> Ga <sub>1-x</sub> N			
$E_{g}(x) =$	$xE_{g}(\ln N) + (1-x)E_{g}(GaN) - 1.4x(1-x)(eV)$		
$\mathbf{P}^{\mathrm{SP}}(x)$	$= [-0.042x - 0.034(1 - x) + 0.037x(1 - x)]C/m^{2}$		
$P_{In_rGa_{t-}}^{Pz}$	$N/\ln N(x) = [-0.113(1 - x) - 0.0276x(1 - x)] C/m^2$		
$P_{In_xGa_{1-}}^{Pz}$	$_{N/GaN}(x) = [0.148x - 0.0424x(1 - x)]C/m^{2}$		
$P^{P_Z}_{In_xGa_1\_}$	$_{N/AIN}(x) = [0.182x - 0.026(1 - x) - 0.0456x(1 - x)] C/m^2$		
Al <sub>x</sub> Ga <sub>1-x</sub>	₹/		
$E_{g}(x) =$	$xE_g(AIN) + (1-x)E_g(GaN) - 0.7x(1-x)(eV)$		
$\mathbf{P}^{\mathrm{SP}}(x)$	$= [-0.090x - 0.034(1 - x) + 0.021x(1 - x)]C/m^{2}$		
$P^{P_Z}_{Al_xGa_1}$	$_{xN/\ln N} = [-0.28x - 0.113(1 - x) + 0.42x(1 - x)] C/m^2$		
$P^{Pz}_{Al_xGa_1}$	$_{xN/GaN} = [-0.0525x + 0.0282x(1 - x)]C/m^2$		
P <sub>Al,Ga1</sub>	$_{\rm N/AIN} = [0.026(1-x) - 0.0248x(1-x)] C/m^2$		
$Al_x In_{1-x} N$			
$E_g(x) =$	$xE_g(AIN) + (1-x)E_g(InN) - 2.5x(1-x)(eV)$		
$P^{SP}(x) =$	$= [-0.090x - 0.042(1 - x) + 0.070x(1 - x)]C/m^{2}$		
$P^{P_{2}}_{Al_{x}In_{1-x}}$	$N/\ln x(x) = [-0.28x + 0.104x(1 - x)] C/m^2$		
$\mathbf{P}_{Al_{z}ln_{1-z}}^{P_{Z}}$	N/GaN(x) = [-0.0525x + 0.148(1 - x) + 0.0938x(1 - x)]C/		
P <sup>Pz</sup>	$N_{AIN} = [0.182(1-x) + 0.092x(1-x)] C/m^2$		

Parameters	GaN	AIN	InN
C33 (GPa)	398	373	224
C44 (GPa)	105	116	48
$d_{13} ({\rm pm/V})$	-1.6	-2.1	-3.5
$d_{33} ({\rm pm/V})$	3.1	5.4	7.6
$d_{15} ({\rm pm/V})$	3.1	3.6	5.5
$P_{sn}$ (C/m <sup>2</sup> )	-0.034	-0.090	-0.042



#### Piezoelectric Polarization in III-Nitrides





### Polarization Engineering in III-Nitride Heterostructures



- Lattice matched heterostructures same lattice constants
- Polarization-matched heterostructures same polarization



#### Semiconductors for Photonics

-OAIP

GaP

AlAs

-AlSb

GaAs

-GaSb

-InAs

GaAs AIAs

temperature (adopted from Tien, 1988).

5.6

5.7

InSb

GaP AIP

5.5

-InP

2.4

2.0

1.6

1.2

0.8

0.4

0.0

5.4

Bandgap energy  $E_g$  (eV)



6.0

10

6.5

InSb

6.4

 GaN-based (InGaN/GaN and AIGaN/GaN) (typically pseudomorphic)

AIN

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GaSb AISb

6.1

6.2

6.3

InAs

6.0

PP P

5.8

5.9

Lattice constant  $a_0$  (Å)

Fig. 7.6. Bandgap energy and lattice constant of various III-V semiconductors at room



200

300

400

500

600

700 800

1000

3.7

UV

Violet Blue

Green

Yellow

Red

IR

InN

3.6

3.5

(um)

2

Wavelength

III-V nitrides T = 300 K $(c \sim 1.6 a_0)$ 

### **Optical absorption in Compound Semiconductors**





Absorption power is Quantified by "Absorption Coefficient"
A <u>negative</u> absorption coefficient is equivalent to <u>optical gain</u>!

• Adapted from: Wolfe/Holonyak/Stillman Physical Properties of Semiconductors

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### **Compound Semiconductor Heterojunction LEDs**



Debdeep Jena (djena@cornell.edu), Cornell University

# GaN based Visible LEDs and Lasers







The Nobel Prize in Physics 2014 Isamu Akasaki, Hiroshi Amano, Shuji Nakamura

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# The Nobel Prize in Physics 2014



Photo: A. Mahmoud Isamu Akasaki Prize share: 1/3



Photo: A. Mahmoud Hiroshi Amano Prize share: 1/3



Photo: A. Mahmoud Shuji Nakamura Prize share: 1/3

The Nobel Prize in Physics 2014 was awarded jointly to Isamu Akasaki, Hiroshi Amano and Shuji Nakamura *"for the invention of efficient blue light-emitting diodes which has enabled bright and energy-saving white light sources"*.

#### Science of light

#### Debdeep Jena (djena@cornell.edu), Cornell University

# How to make a Semiconductor Light Emitter

REVIEWS OF MODERN PHYSICS, VOLUME 87, OCTOBER-DECEMBER 2015

# Nobel Lecture: Background story of the invention of efficient blue InGaN light emitting diodes<sup>\*</sup>

Shuji Nakamura University of California, Santa Barbara, California, USA (published 5 October 2015)





FIG. 2 (color). (a) Image of a blue GaN LED with attached gold wire contacts (size of diode  $0.4 \times 0.4 \text{ mm}^2$ ) and (b) the same LED packaged as a commercial product. From Nakamura, Mukai, and Senoh, 1994.



# How to make white light with semiconductors





# Solid state lighting: The lighting of the future





### Effect of Defects and Quantum Wells on Efficiency



FIG. 13 (color). Dependence of LED efficiency on dislocation density for various semiconductor materials. From Lester *et al.*, 1995, Chichibu *et al.*, 1996, and Nakamura, 1998.

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### Absorption Coefficient of Compound Semiconductors



Number of photons incident per unit area per second:

Vector potential:
$$\mathbf{A}(\mathbf{r}, \mathbf{t}) = \hat{e}A_0 \cos(\mathbf{k_{op}} \cdot \mathbf{r} - \omega t)$$
Poynting Vector: $\mathbf{S}(\mathbf{r}, t) = \mathbf{E}(\mathbf{r}, t) \times \mathbf{H}(\mathbf{r}, t)$  $\mathbf{V}$  $\mathbf{E}(\mathbf{r}, t) = -\frac{\partial}{\partial t}\mathbf{A}(\mathbf{r}, t)$  $= \mathbf{k_{op}}\frac{\omega A_0^2}{\mu}\sin^2(\mathbf{k_{op}} \cdot \mathbf{r} - \omega t)$  $\mathbf{V}$  $\mathbf{E}(\mathbf{r}, t) = -\frac{\partial}{\partial t}\mathbf{A}(\mathbf{r}, t)$  $= -\hat{e}\omega A_0\sin(\mathbf{k_{op}} \cdot \mathbf{r} - \omega t)$  $\mathbf{M}$  $\mathbf{H}(\mathbf{r}, t) = \frac{1}{\mu}\nabla \times \mathbf{A}(\mathbf{r}, t)$  $|\mathbf{G}(\mathbf{r}, t)\rangle = \frac{\omega A_0^2}{2\mu}\mathbf{k_{op}}$  $\mathbf{M}$  $\mathbf{H}(\mathbf{r}, t) = \frac{1}{\mu}\nabla \times \mathbf{A}(\mathbf{r}, t)$  $|\mathbf{G}(\mathbf{r}, t)\rangle| = \frac{\omega A_0^2}{2\mu}\mathbf{k_{op}} = \frac{n_r c\epsilon_0 \omega^2 A_0^2}{2}$ 

Incident photon number per unit area per second: |Poynting vector|/ photon energy

• Goes as square of the amplitude of vector potential (or electric field, or magnetic field)

#### Absorption Spectra of Compound Semiconductors

$$E_{c}(\mathbf{k}) = E_{g} + \frac{\hbar^{2}k^{2}}{2m_{h}^{*}}$$

$$E_{v}(\mathbf{k}) = -\frac{\hbar^{2}k^{2}}{2m_{h}^{*}}$$

$$E_{v}(\mathbf{k}) = E_{g} + \frac{\hbar^{2}k^{2}}{2m_{h}^{*}}$$

$$\frac{1}{m_{r}^{*}} = \frac{1}{m_{e}^{*}} + \frac{1}{m_{h}^{*}}$$

$$\frac{1}{m_{r}^{*}} = \frac{1}{m_{e}^{*}} + \frac{1}{m_{h}^{*}}$$

$$regree 9.5 (a) Optical absorption in a direct-band-gap semiconductor (b) The absorption spectrum due to the interband transitions
$$\alpha_{0}(\hbar\omega) = C_{0}|\hat{e} \cdot \mathbf{p_{cv}}|^{2} \times (g_{s}g_{v}) \times \int_{\mathbf{k}} \frac{d^{3}\mathbf{k}}{(2\pi)^{3}} \delta[E_{c}(\mathbf{k}) - (E_{v}(\mathbf{k}) + \hbar\omega)]$$

$$\alpha_{0}(\hbar\omega) = C_{0}|\hat{e} \cdot \mathbf{p_{cv}}|^{2} \times (g_{s}g_{v}) \times \int_{\mathbf{k}} \frac{d^{3}\mathbf{k}}{(2\pi)^{3}} \delta[E_{c}(\mathbf{k}) - (E_{v}(\mathbf{k}) + \hbar\omega)]$$

$$\sigma_{in}(\hbar\omega) = C_{0}|\hat{e} \cdot \mathbf{p_{cv}}|^{2} \cdot \rho_{r}(\hbar\omega - E_{g})$$

$$\rho_{r}(u) = g_{s}g_{v} \cdot \frac{1}{(2\pi)^{2}} \cdot (\frac{2m_{r}^{*}}{\hbar^{2}})^{\frac{3}{2}} \cdot \sqrt{u}$$

$$C_{0} = \frac{\pi e^{2}}{n_{r}c\varepsilon_{0}m_{0}^{2}\omega}$$

$$This and next few slides:From: S. L. Chuang (Photonic Devices)$$$$

Debdeep Jena (djena@cornell.edu), Cornell University

#### **Optical Matrix Elements for Transitions**



Momentum matrix elements for bulk & quantum well structures

$$M_b^2 = \frac{1}{3} P_x^2 = \frac{m_0^2}{3\hbar^2} P^2$$
$$= \left(\frac{m_0}{m_e^*} - 1\right) \frac{m_0 E_g(E_g + \Delta)}{6(E_g + \frac{2}{3}\Delta)}$$

Table 9.1Summary of the Momentum Matrix Elements in ParabolicBand Model ( $|\hat{e} \cdot \mathbf{p}_{ev}|^2 = |\hat{e} \cdot \mathbf{M}|^2$ )

**Bulk** 
$$|\hat{x} \cdot \mathbf{p}_{cv}|^2 = |\hat{y} \cdot \mathbf{p}_{cv}|^2 = |\hat{z} \cdot \mathbf{p}_{cv}|^2 = M_b^2 = \frac{m_0}{6}E_p$$

Quantum Well

**TE Polarization** 
$$(\hat{e} = \hat{x} \text{ or } \hat{y})$$
  
 $\langle |\hat{e} \cdot \mathbf{M}_{c-hh}|^2 \rangle = \frac{3}{4}(1 + \cos^2 \theta)M_b^2$   
 $\langle |\hat{e} \cdot \mathbf{M}_{c-hh}|^2 \rangle = (\frac{5}{4} - \frac{3}{4}\cos^2 \theta)M_b^2$   
 $\langle |\hat{e} \cdot \mathbf{M}_{c-hh}|^2 \rangle = \frac{3}{2}\sin^2 \theta M_b^2$   
 $\langle |\hat{e} \cdot \mathbf{M}_{c-hh}|^2 \rangle = \frac{1}{2}(1 + 3\cos^2 \theta)M_b^2$   
**Conservation Rule**  
 $\langle |\hat{x} \cdot \mathbf{M}_{c-h}|^2 \rangle + \langle |\hat{y} \cdot \mathbf{M}_{c-h}|^2 \rangle + \langle |\hat{z} \cdot \mathbf{M}_{c-h}|^2 \rangle = 3M_b^2, (h = hh \text{ or } lh)$   
 $\langle |\hat{e} \cdot \mathbf{M}_{c-hh}|^2 \rangle + \langle |\hat{e} \cdot \mathbf{M}_{c-lh}|^2 \rangle = 2M_b^2$ 

### **Optical Gain in Semiconductors**

- We have looked at light absorption by a semiconductor (useful for photodetectors & solar cells)
- But LEDs and LASERs are <u>electrically injected light emitters</u>
- The same theory that explains absorption explains emission under electrical injection as well




### **Optical Gain in Non-Equilibrium Conditions**



#### Absorption Coefficient/Optical Gain in Quantum Wells

$$\begin{split} \psi_{v}(r) &= u_{v}(\mathbf{r}) \times \left[\frac{1}{\sqrt{A}}e^{i\mathbf{k_{t}}\cdot\rho}\right] \times V(n_{v},z) \\ \psi_{c}(r) &= u_{c}(\mathbf{r}) \times \left[\frac{1}{\sqrt{A}}e^{i\mathbf{k'_{t}}\cdot\rho}\right] \times C(n_{c},z) \\ \mathbf{p_{ba}} &= \langle \psi_{c}|\mathbf{p}|\psi_{v}\rangle \approx \langle u_{c}|\mathbf{p}|u_{v}\rangle \times \delta_{\mathbf{k_{t}},\mathbf{k'_{t}}} \times I_{v,n_{v}}^{c,n_{c}} \\ I_{hm}^{en} &= \int_{\infty}^{+\infty} dz C^{\star}(n_{c},z)V(n_{v},z) \end{split}$$
 Determines selection rules

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#### **Optical Gain in Quantum Wells**



Figure 9.7. The stepwise absorption spectrum for a quantum-well structure.



**Figure 9.8.** (a) Population inversion in a quantum well such that  $F_c - F_i > h\omega > E_k + E_{e1} - E_{h1}$ . Here  $F_c$  is measured from the valence band edge where the energy level is chosen to be zero (b) The products of the density of states and the occupation probability for electrons in the conduction band  $\rho_e(E)f_c(E)$  and holes in the valence band  $\rho_b(E)f_b(E) = \rho_b(E)[1 - f_c(E)]$  are plotted vs the energy E in the vertical scale.



#### Measured Gain Spectra in III-V Quantum Wells



Gain spectrum in QWs follows the equilibrium JDOS modulated by the Fermi Dirac functions in accordance with the Bernard-Duraffourg condition.







#### **Compound Semiconductor Laser Diodes**



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### Interband and Intersubband Optical Transitions in QWs

*Using <u>effective-mass theory</u>, optical matrix elements in quantum wells:* 

$$M \propto |\langle f | \mathbf{r} \cdot \eta | i \rangle| = \int \chi_e(z) e^{i\mathbf{k}_{e\perp} \cdot \mathbf{r}_{\perp}} u_{ck_e}(\mathbf{r}) \eta \cdot \mathbf{r} \chi_h(z) e^{i\mathbf{k}_{h\perp} \cdot \mathbf{r}_{\perp}} u_{vk_h}(\mathbf{r}) d\mathbf{r}$$

For **INTERBAND** transitions:

$$M \simeq \sum_{R_i} \chi_e(\mathbf{R_i}) \, \chi_n(\mathbf{R_i}) \, e^{i(\mathbf{k_{h\perp}} - \mathbf{k_{e\perp}}) \cdot \mathbf{R_i}} \int_{\text{cell}} u_{c\mathbf{k_e}}(\mathbf{r}) \, \boldsymbol{\eta} \cdot \mathbf{r} u_{vk_h}(\mathbf{r}) \, d\mathbf{r}$$

 $M_{interband} \sim a_0$ , dipole length ~ lattice constant

For **INTERSUBBAND** (intra-band) transitions:

$$M \sim \int_{\text{crystal}} \chi_e(z) \boldsymbol{\eta} \cdot \mathbf{r} \chi'_e(z) \, dr \int_{\text{cell}} u_{cke}(\mathbf{r}) u^*_{cke}(\mathbf{r}) \, d^3r$$

 $M_{intersubband} \sim L_{well}$ , dipole length  $\sim$  Quantum Well Width (giant dipole effect!!)

# 

(c) INTRABAND TRANSITION  $1e \rightarrow 2e$ 



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#### Intersubband Optical Transitions



Figure 9.11. (a) A simple quantum well with a small doping concentration. (b) A modulationdoped quantum well with a significant amount of screening due to a large doping concentration. (c) The subband energy diagram in the  $k_i$  space. A direct vertical transition occurs because of the k-selection rule in the plane of quantum wells.

 $\psi_{a}(r) = u_{c}(\mathbf{r}) \times \left[\frac{1}{\sqrt{A}}e^{i\mathbf{k_{t}}\cdot\rho}\right] \times \phi_{1}(z)$   $\psi_{b}(r) = u_{c}(\mathbf{r}) \times \left[\frac{1}{\sqrt{A}}e^{i\mathbf{k_{t}}\cdot\rho}\right] \times \phi_{2}(z)$   $\psi_{b}(r) = u_{c}(\mathbf{r}) \times \left[\frac{1}{\sqrt{A}}e^{i\mathbf{k_{t}}\cdot\rho}\right] \times \phi_{2}(z)$   $\mu_{ba} = \langle\psi_{b}|e\mathbf{e}|\psi_{a}\rangle \approx \langle u_{c}|u_{c}\rangle \times \langle\frac{1}{\sqrt{A}}e^{i\mathbf{k_{t}}\cdot\rho}|e\mathbf{r}|\frac{1}{\sqrt{A}}e^{i\mathbf{k_{t}}\cdot\rho}\rangle$   $\approx \delta_{\mathbf{k_{t}},\mathbf{k_{t}'}}\langle\phi_{2}|ez|\phi_{1}\rangle\hat{z} \approx \delta_{\mathbf{k_{t}},\mathbf{k_{t}'}} \times e\mathbf{z_{12}}$ 

 $M_{intersubband} \sim L_{well}$ , dipole length ~ Quantum Well Width (giant dipole effect!!)



#### Intersubband Optical Transitions

(a) Bound-to-bound transition



$$\alpha(\hbar\omega) = \left(\frac{\omega}{n_r c \epsilon_0}\right) \frac{|\mu_{21}|^2 (\Gamma/2)}{(E_2 - E_1 - \hbar\omega)^2 + (\Gamma/2)^2} \left(\frac{m_e^* k_B T}{\pi \hbar^2 L}\right) \ln\left(\frac{1 + e^{(E_F - E_1)/k_B T}}{1 + e^{(E_F - E_2)/k_B T}}\right)$$



**Figure 9.15.** A period of a quantum cascade laser [27] using intersubband transition between  $E_3$  and  $F_2$ . The barriers are Al<sub>0.48</sub> In<sub>0.52</sub> As and the wells are In<sub>0.47</sub> Ga<sub>0.53</sub> As materials. The calculated values are  $E_3 - E_2 = 295$  meV and  $E_3 - E_1 = 30$  meV.

Applications of ISB transitions: In Quantum Cascade Lasers



## **Compound Semiconductor Heterojunction Devices**



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## **Compound Semiconductor Heterojunction Devices**



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### Applications: The Double-Heterostructure Laser

#### DOUBLE-HETEROSTRUCTURE LASER





#### Compound Semiconductor Laser Designs



FIG. 91. Schematics of the gain formation in DH lasers (top) and QW lasers (bottom).



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#### Compound Semiconductor Laser Designs



FIG. 100a. Schematics of band-filling for conduction and valence band states for equal numbers of injected electrons and holes in unstrained (left) and strained (right) QW active layers.



FIG. 113. Schematics of gain curves in 2D, 1D, and 0D structures. Similar numbers of electrons and holes are being injected above transparency, yielding equal integrated gain. (From Nagle and Weisbuch.<sup>465</sup>)

When  $L \gg a_{\rm B}$ , (but still well-separated, confined energy levels), a "giant" oscillator strength situation develops, which yields a transition matrix element<sup>575-7</sup>

$$f \approx f_{\rm at} \cdot (V_{\rm box}/V_{\rm exc}) \tag{112}$$

where  $V_{box}$  and  $V_{exc}$  are the QB and exciton volume, respectively. The enhancement of the oscillator strength originates in the coherent excitation of the QB volume (somewhat analogous to the intersubband giant dipole matrix element of Eq. (54e)), which yields an increased dipole moment.





(Weisbuch/Vinter)

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# Reduction of lasing threshold current density





## Lasers outperform LEDs in efficiency!



FIG. 17 (color). Comparison of external quantum efficiency (EQE) of a commercial LED and laser with increasing current density. From Pourhashemi *et al.*, 2013.

# Lasers outperform LEDs in efficiency!





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Science and Future



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Materials Science of Compound Semiconductors





*Figure 4.1:* A schematic diagram of the periodic arrangement of atoms in a lattice. The basis vectors of the lattice are  $a_x$  and  $a_y$ . Any lattice point can be written as a linear combination of the two basis vectors.



The five fundamental two-dimensional Bravais lattices: 1 – oblique, 2 – rectangular, 3 – centered rectangular, 4 –  $^{6}$  hexagonal (rhombic), and 5 – square. In addition to the stated conditions, the centered rectangular lattice fulfills  $2a_2 - a_1 \perp a_1$ . This orthogonality condition leads to the rectangular pattern indicated and implies  $\varphi \neq 90^\circ$ .

Five Bravais lattices in 2-dimensions







*Figure 4.1:* A schematic diagram of the periodic arrangement of atoms in a lattice. The basis vectors of the lattice are  $a_x$  and  $a_y$ . Any lattice point can be written as a linear combination of the two basis vectors.



*Figure 4.3:* Shows the crystal structures of three compounds used in microelectronics. Each has a different Bravais lattice. To see the Bravais lattices of each compound, look only at the aluminum, strontium, and gallium atoms, respectively.



*Figure 4.2:* Shows the Miller indexing scheme for crystal planes and directions in a simple cubic crystal. Similar indexing methods are used in other Bravais lattices, although the basis vectors are not necessarily parallel to Cartesian coordinate axes. In hexagonal lattices an alternate labeling scheme employing four indices in which only three of the four are independent is often used. Notice that the two  $(\overline{1}\ \overline{1}1)$  planes marked are adjacent and parallel to one another.



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*Figure 4.4:* A comparison of fcc and hcp structures and their corresponding simple structures. Atoms in the "A" plane correspond to the darkest gray circles in the top view while atoms in the "B" and "C" planes are progressively lighter shades of gray.



Figure 4.6: Shows the relationships among the incident,  $\mathbf{k}_i$ , and scattered,  $\mathbf{k}_o$ , wave vectors, the change in wave vector,  $\Delta \mathbf{k}$ , the extra path lengths,  $x_i$  and  $x_o$ , of one part of the scattered wave front along the incident and outgoing waves, and the distance between two scattering centers (atoms) in real space, **R**. Also shown is the equivalent reciprocal space construction based on the Ewald sphere concept showing that diffraction occurs when  $\Delta \mathbf{k}$  begins and ends on a reciprocal lattice point.



Figure 4.5: Shows the effect of crystal size on sensitivity to angular distortions. A "large" crystal shows an obvious effect of a  $2^{\circ}$  twist of one layer with respect to another. By contrast, a "small" crystal shows an almost undetectable effect. To achieve a similarly visible distortion in the small crystal a  $6^{\circ}$  twist is required. The sensitivity to variations in angle produces similar effects on diffraction patterns. Large crystals produce diffraction (constructive interference) only over a very small angular range. Small crystals allow diffracted intensity over a much larger range.

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$$\mathbf{b}_1 = (\mathbf{a}_2 \times \mathbf{a}_3) / (\mathbf{a}_1 \bullet \mathbf{a}_2 \times \mathbf{a}_3)$$
$$\mathbf{b}_2 = (\mathbf{a}_3 \times \mathbf{a}_1) / (\mathbf{a}_1 \bullet \mathbf{a}_2 \times \mathbf{a}_3),$$
$$\mathbf{b}_3 = (\mathbf{a}_1 \times \mathbf{a}_2) / (\mathbf{a}_1 \bullet \mathbf{a}_2 \times \mathbf{a}_3)$$

The reciprocal lattice is defined by the set of points  $hb_1$ ,  $kb_2$ , and  $lb_3$  with the basis vectors defined by Equation 4.2.

$$\frac{1}{d^2} = \frac{h^2 + k^2 + l^2}{a^2}$$



#### Compound Semiconductor Crystal Growth





*Figure 11.2:* Open source evaporation boats (lower) and wire coils (upper figure). Wire coils are generally made of W although other low vapor pressure systems also work. Open boats are made in a wide variety of shapes and of a wide variety of materials including Ta, W, graphite, and other materials.



*Figure 11.1:* Typical evaporation system geometries. Both schematics are simplified and leave out typical components. However, the major parts are indicated. The major difference for the molecular beam epitaxy system is the increased emphasis on ultimate vacuum performance, often four to five orders of magnitude better.

*Figure 11.3:* A simplified schematic sketch of a typical effusion cell. Omitted are the top cap that protects the heater windings from the evaporant as well as improving the black body environment of the crucible, the support clamps and details of the power feedthroughs, etc. The thermocouple junction is pressed gently against the base of the crucible but the black body environment assures an accurate temperature measurement. Note that the heater coils are more closely spaced near the front of the crucible to compensate for greater heat loss in that portion of the cell.



#### Molecular Beam Sources



*Figure 11.4:* The flux distribution of atoms emitted by an effusion cell as a function of the depth in the cell of the evaporant surface relative to the diameter of the surface. [1]. Reprinted from Prog. Crystal Growth and Characterization, Vol 2, Luscher, P.E. and Collins, D.M.; "Design considerations for molecular beam epitaxy systems", pp 15-32, Copyright 1979, with permission from Elsevier.



*Figure 11.6:* Cracking effusion sources for either a solid/liquid source (top) and for a gas source (bottom). The only difference is how the vapor to be cracked is supplied to the baffle section. Normal crackers typically have three baffles.

*Figure 11.5:* The flux patterns resulting from tapered effusion cells in both polar plots (center) and in flux contour plots (right). An isometric schematic shows the design of the tapered effusion cell with a flanged top to reduce creep of effusant out of the cell. Figure redrawn with permission from portions of Figure 3 in Tatuya Yamashita, Takashi Tomita, and Takeshi Sakurai, Japanese Journal of Applied Physics vol. 26 (1987), pages 1192-3. Copywrite 1987, Institute of Pure and Applied Physics.

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#### Materials Sources and Detectors



*Figure 11.7:* A schematic cross section of a typical electron beam evaporator. The magnetic field poles and consequently the magnetic field is out of the plane of the drawing. The high negative potential of the filament assists in electron emission and establishes the electron energy and point of impact on the crucible. Generally electrostatic deflection plates are added to sweep the electron beam into and out of the plane of the drawing.



Figure 11.8: Five methods for monitoring concentrations of gas species. Many others exist.

(Rockett, Chp 11)

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## Tracking crystal growth in-situ using RHEED



*Figure 11.9:* A schematic of the basic apparatus for conducting a RHEED experiment. Not shown is the remainder of the deposition system. This apparatus relies on at least high vacuum to protect the electron gun.

(Rockett, Chp 11)



### Tracking crystal growth in-situ using RHEED



*Figure 11.10:* A schematic of the diffraction patterns in a RHEED experiment resulting from different surface structures. To simplify the drawing the broad rods of the rough surface are drawn as constant in width. Properly, for a two-level surface they have two non-zero Fourier coefficients and should oscillate as sine waves perpendicular to the surface. Therefore the rods should be wider and narrower along their length. See Figure 11.11 for a more accurate representation.

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*Figure 11.11:* Examples of RHEED patterns obtained during non-equilibrium MBE growth of GeSn alloys. (a) Pattern for a very smooth surface as in left side of Figure 11.10 showing points on an arc. The diagonal streaks are multiple diffraction effects known as Kikuchi bands. (b) A pattern for a somewhat rougher surface corresponding to the middle drawing in Figure 11.10. Both (a) and (b) show a half-order reflection between the bright integer-order spots that results from the surface reconstruction. (b) Shows some evidence of residual Kikuchi bands. (c) A pattern for a very rough surface corresponding to the right hand drawing in Figure 11.10 where the electron beam is passing through three dimensional islands. Adapted with permission from O. Gurdal, P. Desjardins, J. R. A. Carlsson, N. Taylor, H. H. Radamson, J.-E. Sundgren, and J. E. Greene, Journal of Applied Physics, 83, 162 (1998). Copyright 1998, American Institute of Physics.

(Rockett, Chp 11)

## Tracking crystal growth in-situ using RHEED



1 monolayer Surface smoothening Normalized RHEED Intensity Envelope decay Growth stopped Growth started Specular beam, off-Bragg 0.0 ż 10 12 Time (sec)

Figure 11.13: A typical example of (hypothetical) RHEED oscillation data during MBE growth at a rate of 1 monolayer/sec. Note that the surface is roughening during the 8 monolayers of deposition as shown by the decay of the oscillations but upon termination of growth the surface becomes smooth and the intensity recovers. To obtain this type of behavior the experiment is configured in the off-Bragg condition so that alternate monolayers of the film interfere destructively. This type of oscillation is observed in the intensity of all reflected and diffracted beams. The most intense is the specularly-reflected beam so that is the one normally used.

*Figure 11.12:* RHEED diffraction Ewald sphere constructions for the on-Bragg and off-Bragg conditions. Note that in the on-Bragg condition there is no difference in the diffracted intensity for the rough and smooth surfaces while for the off-Bragg condition alternate layers of the film interfere destructively. Therefore there is strong contrast for the intensity from rough and smooth surfaces. Refer to Figure 4.6 and related discussion for background on this construction.

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(Rockett, Chp 11)





Figure 4.18: Free energy as calculated from Equations 4.25-4.29. (a) shows the free energy of mixing for several temperatures. Negative values indicate favorable mixing, while positive values indicate a preference for demixing. The more negative the energy the more favorable the mixing. (b) gives the two contributions to the energy, bond enthalpy and heat (temperature x entropy), whose difference gives the curves in (a). Notice in particular that the entropy dependence upon x is significantly more square than the energy dependence. (c) the locus of energy minima from (a) [circles] as a function of temperature determine the boundary of the stable phase regions and the compositions which are stable in the two-phase region. This locus determines the shape of the phase boundaries in the solid phase region of a binary phase diagram. To give an idea of how the rest of the phase diagram enters, the dashed curves are added. Calculation of these requires a separate determination of the energy of the various liquid phases.





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(Rockett, Chps 4 & 6)



phase areas. Numerous In-Se compounds occur which are not labeled to simplify the diagram.

The diagram is approximate and most relevant for 500-750°C.

Figure 4.10: The ternary phase diagram for Al-Ga-As. Each light line represents alloys of constant composition as labeled for each line. The dark horizontal line represents the pseudobinary alloy  $Al_xGa_{1-x}As$ . This is a continuously-soluble alloy as shown in the inset. Figure inset based on data from Reference 1.

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*Figure 4.19:* In a full calculation of a phase diagram one can include the liquid phase. An ideal solution produces a liquid-solid phase diagram as on the left side of the figure. As the bond energy difference from Equation 4.25 becomes more significant the solid tends to prefer to separate into two phases. This has consequences for both the low-temperature behavior described by Equation 4.29 and Figure 4.18 and for the liquid-solid relations. The result is the middle figure. A further increase in the tendency to phase separate leads to the behavior on the right and to a binary eutectic phase diagram as in Figure 4.8.

*Figure 4.20:* A schematic of the process for refining  $SiO_2$  into metallurgical-grade Si using reduction by C. The product CO is removed by evaporation from the surface and is swept away in the gas. Because SiO has a moderate vapor pressure at these temperatures, some SiO also leaves the surface and condenses on cool surfaces nearby.

(Rockett, Chp 4)



*Figure 4.21:* A schematic diagram of the Czochralski crystal growth process. A seed crystal mounted on a cooled chuck is lowered into a molten bath, which will solidify to start crystal growth. This seed is then withdrawn slowly and new crystal grows on the existing seed. If the seed is a single crystal then with proper care the boule will also be a single crystal with zero dislocations. The diameter of the boule is adjusted by control of the shape of the meniscus of the liquid and the shape of the solidification boundary.



*Figure 4.22:* Photographs (left) of a 200-mm Si Czochralski growth facility, and a 300 mm Si boule (right) courtesy of MEMC Electronic Materials, Inc; St. Peters MO. The overlying schematic shows the regions of the boule.



Temperature (K) 1000 700 500 400 350 300 0.1 0.01 0.001 Process Rate (s<sup>-1</sup>) 10 10<sup>-5</sup>  $10^{12} e^{-(2 eV)/k_BT}$ 10<sup>-6</sup> 10 10<sup>-8</sup> 10<sup>-9</sup>  $10^4 e^{-(1 \text{ eV})/k_BT}$ 10-10 10-11 2 3 Inverse Temperature, 1000/T

*Figure 4.23:* Schematic diagram of a typical binary eutectic phase diagram between Si and an impurity near the Si melting point. Insets at the right show the temperature and composition of the system across the solid/liquid interface, as well as a schematic of the interface. The binary phase diagram includes an indication of the compositions of the solid and liquid during growth and the change in both variables through the interface region. As shown, the composition of the liquid is enriched in impurity near the interface due to rejection of extra impurity from the solid during growth. This excess impurity must diffuse into the liquid and results in the composition change shown. Note that the observed segregation ratio will generally be lower during growth than the equilibrium value due to this diffusion rate limitation.

*Figure 4.24:* Rates for two hypothetical processes, one with an activation energy of 2 eV and a prefactor of  $10^{12}$  sec<sup>-1</sup> and a second with an activation energy of 1 eV and a prefactor of  $10^4$  sec<sup>-1</sup>. The high activation energy process is favored by short times at high temperatures (left side of the graph) while the low activation energy process is dominant for long times at low temperatures (right).



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*Figure 6.2:* Shows the ternary phase diagram for Al-Ga-As. There is also significant solubility for Ga in Al (not indicated) and significant vapor pressure for As above a few hundred degrees C. The latter leads to As loss in open or vacuum environments. Inset based on data in Reference [1].



Figure 6.4: The energy gap as a function of lattice constant for II-VI compounds. All of these compounds are direct-gap. The dashed curves are linear interpolations. The real systems would exhibit lattice constant bowing. The black points indicate commonly available substrates on which the II-VI compounds can be grown. Note that HgSe and HgTe are metallic rather than semiconducting. Figure based on References [4] and [5].



*Figure 6.3:* The energy gap of group III-V compounds as a function of lattice constants. The upper plot shows the minimum gaps for the alloys with dashed lines indicating indirect gaps and solid lines for direct gaps. The lower chart shows the minimum direct gaps. Based on data in References [2] and [3]. Note that the nitride values are for the hexagonal wurtzite phase (indicated with a "w"). The cubic nitride lattice constants are 0.438, 0.450, and 0.498 nm for AlN, GaN, and InN, respectively. [3] The cubic nitride energy gaps are similar to or smaller than the hexagonal phase values. Top figure redrawn with permission from I. Vurgaftman, J. R. Meyer, and L. R. Ram-Mohan, Journal of Applied Physics, 89, 5815 (2001). Copyright 2001, American Institute of Physics.



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 $\Delta G = \sum \mu_i \Delta N_i = \Delta E - T \Delta S + P \Delta V,$ 



*Figure 6.5:* The five distinct alloy tetrahedra. White circles indicate A atoms, gray circles B atoms, and black circles C atoms in an  $A_{1-x}B_xC$  type alloy. Note that each tetrahedron can be rotated to place any of the A or B atoms on a given vertex when forming the crystal. The energy of a tetrahedron is independent of its orientation. The model does not consider second-nearest neighbor interactions but can account for bond strains within each tetrahedron. A more accurate model must consider longer-range interactions.

$$\sum_{j} \frac{\partial G}{\partial q_{j}} dq_{j} = 0 \sum_{j} \alpha_{j} \delta q_{j} = 0$$

$$\beta = 1 - \frac{P_{AB}}{x} \qquad \beta = 1 - \frac{q_{1} + 2q_{2}}{x} + \frac{q_{1} + 2q_{2}}{x(1 - x)}$$

$$\Delta \mathbf{E} = 1/2 \left( \mathbf{E}_{\mathrm{AA}} + \mathbf{E}_{\mathrm{BB}} \right) - \mathbf{E}_{\mathrm{AB}},$$
$$\mathbf{E} = N \sum_{j} \alpha_{j} \mathbf{q}_{j} \mathbf{E}_{j},$$

The next step is to calculate the values of  $\alpha_j$ . There are three rotations of the tetrahedra containing mixed A and B atoms that are, to some extent, distinct. The pure AC and pure BC tetrahedra cannot be distinguished upon rotation. Thus, for a diamond or zincblende structure,  $\alpha_j=1$  for the two tetrahedra containing only A and C (*j*=0) or B and C (*j*=4) atoms, while tetrahedra including *j*=1 to 3 B atoms have  $\alpha_j = C_j^4 = 4!/(j!(4-j)!)$  distinguishable rotations. Note that  $\alpha_j q_j$  summed over all five values of j is unity, while the composition, x, of the alloy A<sub>1-x</sub>B<sub>x</sub>C is  $x = q_1+3q_2+3q_3+q_4$ . One may further show that  $q_j=x^j(1-x)^{4-j}$ . [6] Note: the difference between a random and a non-random alloy is in the q<sub>j</sub> values. The ignored third-nearest neighbor values and so this approximation is not normally a problem. Higher order interactions can be important in establishing, for example, the difference between cubic or hexagonal structures and the precise locations of phase boundaries.

$$\boldsymbol{S} = \boldsymbol{k}_{B} N \left( 3 \left[ \boldsymbol{x} \ln \boldsymbol{x} + (1 - \boldsymbol{x}) \ln(1 - \boldsymbol{x}) \right] - \sum_{j} \alpha_{j} \boldsymbol{q}_{j} \ln \boldsymbol{q}_{j} \right)$$

$$\delta x = \delta q_1 + 3\delta q_2 + 3\delta q_3 + \delta q_4 = 0$$

The energy minimization criterion can be used to develop values of the various  $q_j$ 's for given  $E_j$  values. These can, in turn, be used to calculate an order parameter. Consider for a moment the range of values  $\beta$  can take. For a completely phase separated alloy  $P_{AB}=0$  and  $\beta =1$ . If a superstructure order develops in which  $P_{AB}>x$  or  $q_1>x(1-x)^3$ ,  $q_2>x^2(1-x)^2$ , and/or  $q_3>x^3(1-x)$ . This situation would yield  $\beta<0$ . Hence **positive \beta values indicate phase separation while negative \beta shows ordering.** 

(Rockett, Chp 6)

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*Figure 6.6:* Shows the order parameter calculated for  $In_{1-x}Ga_xAs$  as a function of composition at several temperatures. Redrawn with permission from Masaya Ichimura and Akio Sasaki, Journal of Applied Physics, 60, 3850 (1986). Copyright 1986, American Institute of Physics.

For a quaternary  $A_{1-x}B_xC_yD_{1-y}$  alloy, the stability criterion is determined by the second partial derivative of the free energy change,  $\Delta G$ , for mixing:

$$\frac{\partial^2 \Delta G}{\partial x^2} \frac{\partial^2 \Delta G}{\partial y^2} - \frac{\partial^2 \Delta G}{\partial x \partial y} = 0.$$

$$\Delta E = w_{AC}(1-x)(1-y) + w_{BC}x(1-y) + w_{AD}(1-x)y + w_{BD}xy + a_{AC-BC}(1-x)x(1-y) + a_{AD-BD}(1-x)xy + a_{AC-AD}(1-x)(1-y)y + a_{BC-BD}x(1-y)y$$

$$\Delta S = Nk [(1-x)\ln(1-x) + x\ln x + (1-y)\ln(1-y) + y\ln y]$$

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Equilibrium Spinodal Decomposition Phase Boundaries



*Figure 6.7:* Spinodal decomposition curves for six quaternary semiconductor systems as calculated by Onabe in Reference [8]. Crosses indicate the midpoint of the spinodal decomposition regions. Ellipses are equivalent to slices through Figure 4.18c at various temperatures. Figure redrawn with permission from portions of Figure 1 in Kentaro Onabe, Japanese Journal of Applied Physics vol. 21 (1982), pages L323-5. Copyright 1982, Institute of Pure and Applied Physics.



Figure 6.8: Schematics of a zincblende alloy of dark and light gray cations and white anions with CuAu and CuPt ordering on (100) and (111) planes, respectively. A common cation alloy has the same behavior.

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(Rockett, Chp 6)



The similar effects giving rise to both bowing and variations in the energy of the alloy tetrahedra in Figure 6.5 lead to an important conclusion. Bowing and solubility are directly linked. Alloys with large bowings should have large regions where spinodal decomposition would favor phase separation. Thus one may expect poor miscibility in alloys with large bowing. Examination of Equation 6.3 and the equations for the  $\alpha_j$  and  $q_j$  terms also shows why bowing should have a maximum at  $x\sim0.5$  and why it would depend roughly on the difference in bond energies between atoms – because this is where the amplitude of core potential fluctuations should be largest relative to the unperturbed lattice.

Figure 6.9: Schematic cross sections through a zincblende lattice along a (110) direction showing the dimerization of atoms on the surface of the material. The white circles represent common anions in an alloy and the large and small circles are the two cations. The medium gray circles represent the substrate cations. Strain resulting from placement of the large atoms in the second layer influences how dimers tilt on the surface. One side of these dimers favors the larger atom, which continues the underlying order.

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For the virtual crystal of an AB semiconductor alloy the symmetric potential term is

$$\boldsymbol{U}_{s} = \sum_{\boldsymbol{r}'} [\boldsymbol{x} \boldsymbol{U}_{A}(\boldsymbol{r}') + (1 - \boldsymbol{x}) \boldsymbol{U}_{B}(\boldsymbol{r}')].$$

The asymmetric part is best written as

$$\boldsymbol{U}_{a} = \sum_{\boldsymbol{\tau}} [\boldsymbol{U}_{A}(\boldsymbol{r}'-\boldsymbol{\tau}) - \boldsymbol{U}_{B}(\boldsymbol{r}'-\boldsymbol{\tau})]\boldsymbol{c}_{\tau}.$$

$$V_2 [A_{1-x}B_xC \text{ alloy}] = V_2^{AC} + (V_2^{BC} - V_2^{AC}) x_2^{AC}$$



Figure 6.10: Minimum transition energy values for three primary gaps in AlGaAs alloys as a function of alloy composition. Note the direct-indirect transition at approximately 43% Al. Redrawn with permission from I. Vurgaftman, J. R. Meyer, and L. R. Ram-Mohan, Journal of Applied Physics, 89, 5815 (2001). Copyright 2001, American Institute of Physics.

Alloy	Bowing Parameter (eV)			Alloy	Bowing	Parame	X         L           0.22         0.22           0.28         0.28           2.7         2.7		
	Г	X	L		Г	X	L		
AlGaAs	-0.13	0.55	0	AIAsP	0.22	0.22	0.22		
AllnSb	0.43			AIAsSb	0.8	0.28	0.28		
AllnP	-0.48	0.38		AIPSb	2.7	2.7	2.7		
GalnAs	0.48	1.4	0.33	GaAsP	0.19	0.24	0.16		
GalnSb	0.41	0.33	0.4	GaAsSb	1.43	1.2	1.2		
GaAIN	1.0			InAsSb	0.67	0.6	0.6		
GalnN	2.4			GaAsN	120				
All values are for the zincblende structure of the alloy. Most values and especially									
GaAsN, and AlGaAs and AlInP Γ-point values are composition dependent. Data are from									
Reference [2].									

Table 6.1: Semiconductor Alloy Bowing Parameters

$$E_{ABCD}(x,y) = \frac{x(1-x)[(1-y)E'_{ABD}(x) + yE'_{ABC}(x)]}{x(1-x) + y(1-y)} + \frac{y(1-y)[xE'_{ACD}(y) + (1-x)E'_{BCD}(y)]}{x(1-x) + y(1-y)}.$$

(Rockett, Chp 6)

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*Figure 6.14:* A phase diagram for the metastable  $(GaSb)_{1-x}Ge_{2x}$  alloy showing the phases present as a function of deposition temperature and composition based on data in Cadien et al. [20]. The darker gray area is the single-phase alloy region. The light gray region is a mixture of alloy and two-phase regions, and at sufficiently high temperature only the two equilibrium phases are found. Open data points are for films on GaAs (100) single-crystal substrates while the closed points are for films on glass.

*Figure 6.15:* A phase diagram for the metastable  $(GaSb)_{0.9}Ge_{0.1}$  alloy showing the phases present as a function of deposition temperature and fast particle bombardment energy. Figure is based on data in Shah [21]. Circles indicate single-phase metastable alloys formed while square points show two-phase samples.

(Rockett, Chp 6)

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Figure 4.12: The energy of a system of atoms during the course of a reaction such as  $2A + BC_2 \rightarrow A_2B + C_2$ . The reactants must be configured for reaction. In the example this might mean attaching two A atoms to the BC<sub>2</sub> molecule prior to reaction. The reverse reaction would cross a higher energy barrier climbing the entire energy hill from the right ("Products") to the activated state energy.

$$2A + BC_2 \rightarrow A_2B + C_2$$
$$R_f = k_f \frac{[A]^2 [BC_2]}{[A_2B] [C_2]},$$
$$k_f = k_0 e^{-\Delta H/k_B T}$$

A+B
$$\frac{k_{1}}{k_{-1}}$$
A:B $\frac{k_{2}}{k_{-2}}$ C:D $\frac{k_{3}}{k_{-3}}$ C+D  
 $k_{f} = \frac{k_{1}}{1 + (k_{-1}/k_{2}) + (k_{-1}k_{-2}/k_{2}k_{3})}$   
 $k_{r} = \frac{k_{-3}}{1 + (k_{3}/k_{-2}) + (k_{2}k_{3}/k_{-1}k_{-2})}$ 

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$$E = \pi r^2 \gamma_s + (4\pi/3)r^3 \Delta H$$

$$r^* = -\frac{\gamma_s}{2\Delta H}$$

$$\boldsymbol{E}^* = \frac{\pi}{12} \frac{\gamma_s^3}{\Delta \boldsymbol{H}^2}$$

$$R = R_0 e^{-E^*/k_B T}$$



*Figure 4.13:* Shows the energy of a cluster of atoms during nucleation as a function of cluster size. Surface energy generally increases as  $r^2$ , while volume energy decreases as  $r^3$ . The energy of the cluster increases with addition of atoms up to a critical size,  $r^*$ , beyond which the energy decreases with atom addition. The maximum energy at  $r^*$  and is the activation energy barrier for nucleation.

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*Figure 4.14:* A schematic of the atomic movements involved in a typical substitutional diffusion process in a fcc lattice and the change in system energy throughout the process.

 $D = D_0 e^{-E_D/kT}$ 

$$F = -D\frac{dC}{dx}$$

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#### Next topic: Defects in Semiconductors

#### Rockett: MSS

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**Figure 2.1** Defects in GaAs. Gallium vacancy ( $V_{Ga}$ ), arsenic interstitial (As<sub>*i*</sub>), arsenic antisite (As<sub>Ga</sub>), and substitutional carbon ( $C_{As}$ ).

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Materials Science of Compound Semiconductors

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## Interstitial sites in FCC lattices, Electronic nature





<b>Defect Property</b>	Traps	<b>Recombination Centers</b>
Charge state before	Doubly charged,	Singly charged, opposite
capture of a minority	opposite to minority	to minority carriers
carrier:	carriers	
Cross section for minority	$10^{-9}$ to $10^{-12}$ cm <sup>2</sup>	$10^{-12}$ to $10^{-15}$ cm <sup>2</sup>
carriers before capture:	(strongly attractive)	(moderately attractive)
Cross section for majority	$10^{-18}$ - $10^{-20}$ cm <sup>2</sup>	$10^{-15}$ to $10^{-17}$ cm <sup>2</sup>
carriers after capture:	(moderately repulsive)	(neutral)
Depth (energy from	Relatively small	Relatively large
minority carrier band		
edge):		

Table 7.1: Ideal Properties for Traps and Recombination Centers



hole

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## Electronic Energy levels of Substitutional Dopants



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## Electronic Energy levels of substitutional impurities



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## Electronic Energy levels of Hydrogen Interstitial



**Figure 4.11** Valence band maxima and conduction band minima (solid lines), and the H(+/–) level (dashed line), as calculated by density functional theory (DFT) for various semiconductors. Here, zero energy corresponds to the vacuum level. (After Van de Walle, C.G., and Neugebauer, J., Universal Alignment of Hydrogen Levels in Semiconductors, Insulators, and Solutions, *Nature*, 423, 626–628, 2003.)

(Rockett, Chp 7, Haller/McLuskey)



#### Formation Energy & Statistics of Vacancies & Interstitials



To illustrate the effect of Equations 7.6 and 7.7 on the vacancy concentration, let us consider the case of undoped and doped Si. Here vacancies can occur in three charge states, +1, -1, and -2, with energies 1.0 eV below the conduction band, and 0.6 and 1.05 eV above the valence band at 300 K, respectively. The overall energy gap of Si is 1.12 eV. The energy of formation of a neutral vacancy is 2.4 eV, the vibrational entropy change per added vacancy is 1.1 k<sub>B</sub>, and the site density in the Si lattice is  $5x10^{22}$  cm<sup>-3</sup>.[3]

$$\Delta E_d(q, E_f) = \Delta E_0(q) - q\Delta E_f$$

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#### Formation Energy & Statistics of Vacancies & Interstitials



**Figure 7.1** Formation enthalpies for native defects in GaAs. The Fermi stabilization energy  $E_{FS}$  is ~0.6 eV above the valence-band maximum. (After Walukiewicz, W., Amphoteric Native Defects in Semiconductors, *Appl. Phys. Lett.*, 54, 2094–2096, 1989.)



**Figure 7.2** Approximate Fermi stabilization energy level  $E_{FS}$ , for various III-V semiconductors. (After Walukiewicz, W., Intrinsic Limitations to the Doping of Wide-Gap Semiconductors, *Physica B*, 302–303: 123–134, 2001. For consistency, the band alignments of Figure 4.11 were used.)

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#### Formation Energy & Statistics of Vacancies & Interstitials



**Figure 7.4** Formation enthalpies for oxygen and zinc vacancies in ZnO, for zinc-rich and oxygen-rich conditions. The Fermi energy is relative to the valence-band maximum. (After Janotti, A., and Van de Walle, C.G., Native Point Defects in ZnO, *Phys. Rev. B*, 76, 165202, 1–22, 2007.)

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#### Strain Energies: Microscopic



Figure 4.1: Valence force field potential energies versus As position within a 5atom  $In_{0.5}Ga_{0.5}As$  tetrahedron. Left and right panels show contributions due to bond bending and bond stretching forces; center panel shows the sum of the two contributions. The geometric center of the tetrahedron is at z = 0; the actual As position is shifted downward toward the Ga atoms. The predictions of the virtual crystal (VCA) and covalent radius (CRA) approximations discussed in the text are also shown.

Material	α	β	$\beta/\alpha$	C <sub>11</sub>	$C_{12}$	$C_{44}$
				$(10^{10})$	$(10^{10})$	$(10^{10})$
	(N/m)	(N/m)		$N/m^2)$	$N/m^2)$	$N/m^2)$
С	129.33	84.76	0.655	107.6	12.50	57.68
Si	48.50	13.81	0.285	16.57	6.39	7.96
Ge	38.67	11.35	0.294	12.89	4.83	6.71
AlSb	35.35	6.77	0.192	8.94	4.43	4.16
GaP	47.32	10.44	0.221	14.12	6.25	7.05
GaAs	41.19	8.95	0.217	11.81	5.32	5.92
GaSb	33.16	7.22	0.218	8.84	4.03	4.32
InP	43.04	6.24	0.145	10.22	5.76	4.60
InAs	35.18	5.50	0.156	8.33	4.53	3.96
InSb	29.61	4.77	0.161	6.67	3.65	3.02
ZnS	44.92	4.78	0.107	10.40	6.50	4.62
ZnSe	35.24	4.23	0.120	8.10	4.88	4.41
ZnTe	31.35	4.45	0.142	7.13	4.07	3.12
CdTe	29.02	2.43	0.084	5.35	3.68	1.99
CuCl	12.60	1.00	0.079	2.72	1.87	1.57

Table 4.1: Microscopic bond stretching  $(\alpha)$  and bond bending  $(\beta)$  force constants deduced from macroscopic elastic constants  $(C_{11}, C_{12}, \text{ and } C_{44})$  of various cubic semiconducting materials.<sup>*a*</sup>

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$$u_{\mathrm{str},i} = \frac{3}{8} \alpha_i \frac{\left(d_i^2 - d_{i,o}^2\right)^2}{d_{i,o}^2}$$

$$u_{\text{bnd},ij} = \frac{3}{8} \frac{\beta_i + \beta_j}{2} \frac{\left(\mathbf{d}_i \cdot \mathbf{d}_j - \mathbf{d}_{i,\text{o}} \cdot \mathbf{d}_{j,\text{o}}\right)^2}{d_{i,\text{o}} d_{j,\text{o}}}$$

$$u_{\text{str}} = \sum_{i=1}^{4} u_{\text{str},i}$$
  $u_{\text{bnd}} = 2 \sum_{i=1}^{4} \sum_{j < i} u_{\text{bnd},ij}$ 

$$\begin{aligned} u_{(\blacktriangle)} &= u_{\rm str} + u_{\rm bnd} \\ &\approx \frac{3}{4} \left[ \alpha_{\rm GaAs} \frac{\left(d_{\rm GaAs}^2 - d_{\rm GaAs,o}^2\right)^2}{d_{\rm GaAs,o}^2} + \alpha_{\rm InAs} \frac{\left(d_{\rm InAs}^2 - d_{\rm InAs,o}^2\right)^2}{d_{\rm InAs,o}^2} \right] \\ &+ \frac{3}{4} \left[ \beta_{\rm GaAs} \frac{\left(d_{\rm GaAs}^2 \cos^2\theta_{\rm GaAsGa} - d_{\rm GaAs,o}^2 \cos^2\theta_{\rm T}\right)^2}{d_{\rm GaAs,o}^2} \right] \\ &+ \beta_{\rm InAs} \frac{\left(d_{\rm InAs}^2 \cos^2\theta_{\rm InAsIn} - d_{\rm InAs,o}^2 \cos^2\theta_{\rm T}\right)^2}{d_{\rm InAs,o}^2} \right]. \end{aligned}$$

(Rockett, Chp 7, Tsao, Marder)

#### Strain Energies: Microscopic



Figure 4.1: Valence force field potential energies versus As position within a 5atom  $In_{0.5}Ga_{0.5}As$  tetrahedron. Left and right panels show contributions due to bond bending and bond stretching forces; center panel shows the sum of the two contributions. The geometric center of the tetrahedron is at z = 0; the actual As position is shifted downward toward the Ga atoms. The predictions of the virtual crystal (VCA) and covalent radius (CRA) approximations discussed in the text are also shown.

Material	α	β	$\beta/\alpha$	C <sub>11</sub>	$C_{12}$	$C_{44}$
				$(10^{10})$	$(10^{10})$	$(10^{10})$
	(N/m)	(N/m)		$N/m^2)$	$N/m^2)$	$N/m^2)$
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Table 4.1: Microscopic bond stretching  $(\alpha)$  and bond bending  $(\beta)$  force constants deduced from macroscopic elastic constants  $(C_{11}, C_{12}, \text{ and } C_{44})$  of various cubic semiconducting materials.<sup>*a*</sup>

 $d_{\text{GaAs}}^{2} = \left(\frac{a_{\text{VC}}}{2\sqrt{2}}\right)^{2} + \left(\frac{a_{\text{VC}}}{4} + z\right)^{2}$  $d_{\text{InAs}}^{2} = \left(\frac{a_{\text{VC}}}{2\sqrt{2}}\right)^{2} + \left(\frac{a_{\text{VC}}}{4} - z\right)^{2}$ 

$$\cos( heta_{
m GaAsGa}/2) ~pprox ~rac{(a_{
m VC}/4)+z}{(\sqrt{3}a_{
m VC}/4)+z/\sqrt{3}} \ \cos( heta_{
m InAsIn}/2) ~pprox ~rac{(a_{
m VC}/4)-z}{(\sqrt{3}a_{
m VC}/4)-z/\sqrt{3}},$$

$$\begin{aligned} u_{(\blacktriangle)} &\approx \frac{3}{2} \alpha_{\rm GaAs} \left[ \frac{\sqrt{3}}{4} a_{\rm VC} - d_{\rm GaAs,o} + \frac{z}{\sqrt{3}} \right]^2 \\ &+ \frac{3}{2} \alpha_{\rm InAs} \left[ \frac{\sqrt{3}}{4} a_{\rm VC} - d_{\rm InAs,o} - \frac{z}{\sqrt{3}} \right]^2 \\ &+ \frac{3}{8} \beta_{\rm GaAs} \left[ \frac{-2}{3} \left( \frac{\sqrt{3}}{4} a_{\rm VC} - d_{\rm GaAs,o} \right) + \frac{2}{\sqrt{3}} z \right]^2 \\ &+ \frac{3}{8} \beta_{\rm InAs} \left[ \frac{-2}{3} \left( \frac{\sqrt{3}}{4} a_{\rm VC} - d_{\rm InAs,o} \right) - \frac{2}{\sqrt{3}} z \right]^2. \end{aligned}$$
(4.9)

Then, solving for  $\partial u_{(\bigstar)}/\partial z = 0$ , the equilibrium position can be deduced to be

$$z_{\rm equ} = \left(\frac{-\sqrt{3}}{2}\right) \frac{\alpha_{\rm GaAs} + \alpha_{\rm InAs} - \beta_{\rm GaAs}/3 + \beta_{\rm InAs}/3}{\alpha_{\rm GaAs} + \alpha_{\rm InAs} - \beta_{\rm GaAs} + \beta_{\rm InAs}} (d_{\rm InAs,o} - d_{\rm GaAs,o}).$$

$$- \qquad (4.10)$$

(Rockett, Chp 7, Tsao, Marder)

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#### Strain Energies: Microscopic



$$u_{(\blacktriangle)} \approx \frac{\overline{\alpha}\overline{\beta}}{2(\overline{\alpha} + \overline{\beta})} (\Delta a_o)^2,$$

$$\Delta a_{
m o} \equiv a_{
m InAs,o} - a_{
m GaAs,o} = rac{4}{\sqrt{3}} (d_{
m InAs,o} - d_{
m GaAs,o})$$





Figure 4.3: Regular solution parameters for various pseudobinary alloys. Values plotted along the bottom axis were calculated using Equation 4.13; values plotted along the left axis are experimental measurements<sup>a</sup>; values plotted along the right axis are the critical temperatures, deduced from Equation 3.38, above which the constituent components are fully miscible.

(Rockett, Chp 7, Tsao, Marder)

 Table 4.2: Crystal structures, room-temperature lattice parameters and thermal expansion coefficients of various semiconductors.<sup>a</sup>

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R

 $T_{misc}$ 

Figure 4.2: Measured and calculated Ga-As and In-As bond lengths in  $In_{1-x}Ga_x$ As alloys. Experimental data are from (open circles) X-ray-absorption fine structure (EXAFS)<sup>a</sup> measurements; calculations are based on valence-force-field potentials using either bond bending forces in the virtual crystal approximation (dashed line), bond stretching forces in the covalent radius approxir (dotted lines), or both (solid lines).<sup>b</sup>

Material	Crystal	$a_{\rm o}$ or $a/c$	$\alpha_T = \partial \ln a_{\rm o} / \partial T$
	Structure	(A  or  A/A)	$(10^{-6} \mathrm{K}^{-1})$
С	Diamond	3.56683	0.87 + 0.0092(T - 273)
Si	Diamond	5.43095	3.08 + 0.0019(T - 273)
Ge	Diamond	5.64613	6.05 + 0.0036(T - 273)
$\alpha$ -Sn	Diamond	6.48920	
SiC	Wurtzite	3.086/15.117	
BN	Zincblende	3.6150	
BP	Zincblende	4.5380	
AlP	Zincblende	5.4510	
AlAs	Zincblende	5.6605	3.40 + 0.0064(T - 273)
AlSb	Zincblende	6.1355	
GaN	Zincblende	3.189/5.185	
GaP	Zincblende	5.4512	5.81
GaAs	Zincblende	5.6533	5.35 + 0.0080(T - 273)
GaSb	Zincblende	6.0959	6.7
InP	Zincblende	5.8686	
InAs	Zincblende	6.0584	4.33 + 0.0038(T - 273)
InSb	Zincblende	6.4794	
ZnO	Rock Salt	4.580	
ZnS	Zincblende	5.420	6.70 + 0.0128(T - 313)
ZnS	Wurtzite	3.82/6.26	
CdS	Zincblende	5.8320	
CdS	Wurtzite	4.16/6.756	
CdTe	Zincblende	6.482	
CdSe	Zincblende	6.050	
PbS (Galena)	Rock Salt	5.9362	18.81 + 0.0074(T - 273)
PbTe (Altaite)	Rock Salt	6.4620	19.80
			· · · · · · · · · · · · · · · · · · ·

$$a_{
m sub} = (1 - x_{
m sub}) a_{
m GaAs,o} + x_{
m sub} a_{
m InAs,o}$$

 $a_{\text{epi},\parallel} = a_{\text{sub}} = (1 - x_{\text{sub}})a_{\text{GaAs,o}} + x_{\text{sub}}a_{\text{InAs,o}}.$ 

 $\epsilon_{\mathrm{epi},\parallel} \equiv 2 \frac{a_{\mathrm{epi},\parallel} - a_{\mathrm{epi},\mathrm{o}}}{a_{\mathrm{epi},\parallel} + a_{\mathrm{epi},\mathrm{o}}}$ 

 $a_{\rm epi,o} = (1 - x_{\rm epi})a_{\rm GaAs,o} + x_{\rm epi}a_{\rm InAs,o}$ 

If the epitaxial film and its substrate are oriented along one of the  $\langle 100 \rangle$  cubic symmetry directions, then this equation reduces to

$$\begin{pmatrix} \sigma_{\rm epi,\parallel} \\ \sigma_{\rm epi,\perp} \end{pmatrix} = \begin{pmatrix} C_{11} + C_{12} & C_{12} \\ 2C_{12} & C_{11} \end{pmatrix} \begin{pmatrix} \epsilon_{\rm epi,\parallel} \\ \epsilon_{\rm epi,\perp} \end{pmatrix}.$$
(4.20)

(Rockett, Chp 7, Tsao, Marder)



$$egin{aligned} \epsilon_{i,\parallel} &=& 2rac{a_{ ext{epi},\parallel}-a_{i, ext{o}}}{a_{ ext{epi},\parallel}+a_{i, ext{o}}} \ \epsilon_{i,\perp} &=& 2rac{a_{ ext{epi},\perp}-a_{i, ext{o}}}{a_{ ext{epi},\perp}+a_{i, ext{o}}} \end{aligned}$$

$$\sigma_{\mathrm{epi},\perp} = 2C_{12}\epsilon_{\mathrm{epi},\parallel} + C_{11}\epsilon_{\mathrm{epi},\perp} = 0$$

$$\begin{split} \epsilon_{\mathrm{epi},\perp} &= \frac{-2C_{12}}{C_{11}} \epsilon_{\mathrm{epi},\parallel} \\ a_{\perp}(x_{\mathrm{epi}}, x_{\mathrm{sub}}) &= a_{\mathrm{epi},\mathrm{o}} \frac{1 + \epsilon_{\mathrm{epi},\perp}/2}{1 - \epsilon_{\mathrm{epi},\perp}/2} \end{split}$$

$$\begin{aligned} u_{i,\text{ext}} &= \frac{1}{2} \left[ 2\sigma_{i,\parallel} \epsilon_{i,\parallel} + \sigma_{i,\perp} \epsilon_{i,\perp} \right] \\ &= \left( C_{i,11} + C_{i,12} \right) \epsilon_{i,\parallel}^2 + 2C_{i,12} \epsilon_{i,\perp} \epsilon_{i,\parallel} + \frac{1}{2} C_{i,11} \epsilon_{i,\perp}^2 \\ \left( \frac{\epsilon_x}{\epsilon_y} \right) &= \frac{1}{2\mu(1+\nu)} \begin{pmatrix} 1 & -\nu & -\nu \\ -\nu & 1 & -\nu \\ -\nu & -\nu & 1 \end{pmatrix} \begin{pmatrix} \sigma_x \\ \sigma_y \\ \sigma_z \end{pmatrix} \qquad \begin{pmatrix} \epsilon_\parallel \\ \epsilon_\perp \end{pmatrix} = \frac{1}{2\mu(1+\nu)} \begin{pmatrix} 1-\nu & -\nu \\ -2\nu & 1 \end{pmatrix} \begin{pmatrix} \sigma_\parallel \\ \sigma_\perp \end{pmatrix} \\ C_{11} &= 2\mu \left( \frac{1-\nu}{1-2\nu} \right) \\ C_{12} &= 2\mu \left( \frac{\nu}{1-2\nu} \right) \end{aligned} \qquad \sigma_\parallel = 2\mu \left( \frac{1+\nu}{1-\nu} \right) \epsilon_\parallel \end{aligned}$$
(Rockett, Chp 7, Tsao, Marder)

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(Rockett, Chp 7, Tsao, Marder)





$$\begin{split} b_{\text{edg},\parallel} &\equiv b \cos \lambda \\ U_{\text{edg}} &= \frac{\mu b_{\text{edg}}^2}{4\pi (1-\nu)} \ln(R/r_o) \\ U_{\text{scr}} &= \frac{\mu b_{\text{scr}}^2}{4\pi} \ln(R/r_o), \\ U_{\text{dis}} &= \frac{\mu b^2}{4\pi} \left(\frac{\sin^2 \beta}{1-\nu} + \cos^2 \beta\right) \ln(4R/b) \\ &= \frac{\mu b^2}{4\pi} \left(\frac{1-\nu \cos^2 \beta}{1-\nu}\right) \ln(4R/b). \end{split}$$

$$U_{\rm dis} = \frac{\mu b^2}{4\pi} \left( \frac{1 - \nu \cos^2 \beta}{1 - \nu} \right) \ln(4h/b).$$



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Figure 5.9: Equilibrium misfit strain  $(f_{dis})$  taken up by dislocations as a function of thickness at constant misfit (left) or as a function of misfit at constant thickness (right). Below either the critical thickness  $(h_c)$  or critical misfit  $(f_c)$  the strain taken up by dislocations is zero; above the critical thickness or critical misfit it becomes an increasingly large fraction of the total misfit (f).

$$\begin{split} \rho_{\rm md,equ} &= \frac{f}{b\cos\lambda} \\ &- \frac{b}{8\pi h\cos^2\lambda} \left(\frac{1-\nu\cos^2\beta}{1+\nu}\right) \ln\left(\frac{4h}{b}\right) = \frac{f}{b\cos\lambda} \left[1-\frac{h_{\rm c}}{h}\right] \end{split}$$

(Rockett, Chp 7, Tsao, Marder)



#### **Dislocations in Buried Heterostructures & Motion**



Figure 5.10: Strain relaxation by single (left) and double (right) kinking of a threading dislocation to form misfit dislocations.



Figure 5.17: Schematic illustrations of possible microscopic deformation mechanisms operative during strain relaxation.



Figure 5.22: Uncorrugated (left) and corrugated (right) epitaxial strained layers.



Figure 5.12: Strains, equivalent strains, and excess stresses in a double buried quantum-well heterostructure. Both the single- and double-kink excess stresses maximize at a depth of 800 Å, but only the single-kink excess stress exceeds zero, and even then just barely. Therefore, this structure is stable with respect to double-kink strain relaxation, but slightly unstable with respect to single-kink strain relaxation.

(Rockett, Chp 7, Tsao, Marder)





**Figure 14.6.** In order to add a new layer of atoms to a crystal, it is typically necessary to add a large clump of new atoms simultaneously. (A) A single atom attaching to the surface will be unstable and detach. (B) A large enough clump is stable.



**Figure 14.7**. Mechanism due to Frank, showing how the presence of a screw dislocation should speed the process of crystal growth, by providing open regions to which atoms can easily attach.



Figure 14.8. Experimental observation of growth spirals on the (0001) plane of silicon carbide. [Source: Amelinckx (1964), p. 5.]

#### (Rockett, Chp 7, Tsao, Marder)







*Figure 7.11:* A schematic diagram of various edge dislocation configurations. (a) a classic edge dislocation with an extra half plane inserted into the material. This type of dislocation forms a loop which closes at the crystal surface. Thus, threading segments will occur. (b) and (c) are loops resulting from coalescence of interstitials and vacancies, respectively to form small planes. When these result in stacking faults, the faults are termed extrinsic and intrinsic for (b) and (c), respectively and have dislocations at their edges.

(Rockett, Chp 7, Tsao, Marder)



#### **Dislocation Energetics and Fields**

$$\Gamma_{screw} = \frac{G}{4\pi} \left[ b^2 \ln\left(\frac{r}{b}\right) \right] + E_{core}$$

$$\sigma_{rr} = -\frac{Gb}{2\pi(1-\nu)r^2l}\vec{r}\cdot\vec{\xi}$$

$$P_{edge} = -\frac{(1+\nu)Gb\sin\theta}{3\pi(1-\nu)r}$$

$$E = \frac{4}{3} \frac{(1+\nu)}{(1-\nu)} \frac{R\Delta R}{r} \sin\theta$$



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a



## **Dislocation Energetics: Critical Thickness**





#### 11% Lattice Mismatch



*Figure 7.16:* A cross-sectional transmission electron micrograph of a GaAs/InSb heterojunction. The mismatch in this system is approximately 14.6%, giving rise to a misfit dislocation roughly every seven lattice spacings. In this image, the core regions of the edge dislocations appear as brighter regions. The schematic below shows a sketch of the dislocation structure in the image above. Each marked dislocation terminates two of the (111)-type lattice fringes as shown and is located in the whiter regions along the marked interface.

(Rockett, Chp 7, Tsao, Marder)



#### **Dislocation Energetics: Critical Thickness**







Figure 7.18: The equilibrium critical thickness for a film with a Poisson's ratio of 0.22 assuming strain relief by 60° dislocations ( $\theta$ ,  $\phi = 60^{\circ}$ ). The curve was calculated using Equation 7.29, with  $\alpha = 1$ .

(Rockett, Chp 7, Tsao, Marder)

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## Dislocation Arrays: Critical Thickness

$$f(D) = f_{\infty} - \frac{b\cos\phi}{D}$$
$$E_{array} = \frac{Gb^2(1 - v\cos^2\theta)}{8\pi(1 - v)D} \ln\left(\frac{\alpha h}{b}\right)$$

$$D(h) = \frac{b\cos\phi}{\frac{\Delta a}{a} - \frac{b(1 - v\cos^2\theta)}{8\pi(1 - v)\cos\phi} \frac{\ln(\alpha h/b)}{h}}$$

$$D(h) = \frac{b\cos\phi}{\frac{\Delta a}{a} - C\frac{\ln(\alpha h/b)}{h}}$$

 $\mathbf{D}=\infty$ 



Figure 7.19: A transmission electron micrograph looking down upon the interface between a GaAs substrate and an InAs epitaxial layer. Both threading segments stretching from the heterojunction to the surface and reactions between dislocation segments are visible in the

micrograph. The lines in the image are the misfit dislocations in the interface. [8]

(Rockett, Chp 7, Tsao, Marder)



## **Dislocation Scattering and Mobility**

APPLIED PHYSICS LETTERS

VOLUME 76, NUMBER 13



#### Dislocation scattering in a two-dimensional electron gas

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FIG. 1. Line dislocation modeled as a line of charge. The dislocation line has filled acceptor states along it. Charges on the dislocation line act as an extended remote impurity. The remote ionized impurity matrix element is integrated for all the small elemental remote impurities to account for the effect of the whole dislocation.



FIG. 3. Dislocation scattering inhibited 2DEG mobility. Mobilities for three different dislocation densities have been shown. Our theory predicts an approximate  $n_s^{3/2}/N_{\rm dis}$  mobility dependence, which is seen in this plot. Note the strong mobility degradation at  $10^{10}$  cm<sup>2</sup> dislocation density.

$$\begin{aligned} \tau_{\rm dis}^{\rm 3D} &= \frac{\hbar^3 \epsilon_0^2 \epsilon_b^2 c_0^2}{N_{\rm dis} m^* e^4 f^2} \cdot \frac{(1 + 4\lambda_D^2 k_\perp^2)^{\frac{3}{2}}}{\lambda_D^4} \propto \frac{(k_B T)^{\frac{3}{2}}}{\lambda_D} \\ \tau_{\rm dis}^{\rm 2D} &= \frac{\hbar^3 \epsilon_0^2 \epsilon_b^2 c_0^2}{N_{\rm dis} m^* e^4 f^2} \cdot \frac{16\pi k_F^4}{I\left(\frac{q_{\rm TF}}{2k_F}\right)} \\ &\approx \frac{\hbar^3 \epsilon_0^2 \epsilon_b^2 c_0^2}{N_{\rm dis} m^* e^4 f^2} \cdot \frac{16\pi k_F^4}{\left(\frac{1.84k_F}{q_{\rm TF}} - 0.25\right)} \end{aligned}$$



## **Dislocation Scattering and Mobility**

APPLIED PHYSICS LETTERS

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#### Effect of scattering by strain fields surrounding edge dislocations on electron transport in two-dimensional electron gases

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(Received 16 July 2001; accepted for publication 20 October 2001)

 $\Delta E_{C} = a_{C}Tr(\epsilon)$ 

$$\delta V = \Delta E_C = a_C Tr(\epsilon) = -\frac{a_C b_e}{2\pi} \frac{1-2\gamma}{1-\gamma} \frac{\sin(\theta)}{r}$$

$$\mu_{\rm disl}^{\rm strain} = \frac{2e\hbar^3 \pi k_F^2}{N_{\rm disl}m^{*2}b_e^2 a_C^2} \left(\frac{1-\gamma}{1-2\gamma}\right)^2 \frac{1}{I(n_s)}$$





FIG. 1. The band electron experiences the depicted CB minimum fluctuation caused by strain fields around an edge dislocation. Strain is anisotropic, with maximum strain in directions perpendicular to the Burgers vector. The energy is in arbitrary units.

#### (Rockett, Chp 7, Tsao, Marder)



## **Dislocation Scattering and Mobility**

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#### Electron Transport in III-V Nitride Two-Dimensional Electron Gases

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Fig. 1. a) Scattering processes limiting electron mobility in the Al(Ga)N/GaN 2DEG system evaluated for  $N_{\text{disl}} = 5 \times 10^8 \text{ cm}^{-2}$ ; experimental data is also shown (circles). b) Effect of dislocation scattering for three different dislocation densities

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## Effect of Dislocations on Vertical Diodes & Transistors

#### Diagram and AFM image of LEO grown template

SEM micrograph of LEO Cross-section





Figure 5.5: The above cross section shows an as-cleaved cross section of uncoalesced LEO stripes similar to those in the LEO template utilized for the AlGaN/GaN HBT on LEO GaN (courtesy P. Fini [9]).

omparison of collector-emitter leakage on wing vs window



Figure 5.7: Semi log plot of leakage current of LEO window compared to wing region. Plot shows reduction of leakage by four orders of magnitude for wing region as compared to window region. The emitter mesa area for these devices was 6  $\mu$ mx 50  $\mu$ m.

Figure 5.6: Atomic Force Microscopy image of an LEO substrate (above) showing window and ring regions. Spiral growth mode in the window region is associated with the screw component of TDs. Wing regions consist of atomically flat steps.

#### Lee McCarthy (PhD Thesis, 2001)


#### Effect of Dislocations on Vertical Diodes & Transistors

APPLIED PHYSICS LETTERS 107, 232101 (2015)

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#### High breakdown single-crystal GaN p-n diodes by molecular beam epitaxy

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TABLE I. Substrates specifications for MBE grown p-n junctions.  $\sigma$ : carrier concentration,  $\mu$ : carrier mobility, *t*: thickness, *TD*: threading dislocation density, and  $\rho$ : resistivity at 300 K.

GaN Substrate	$\sigma$ (cm <sup>-3</sup> )	$\mu$ (cm <sup>2</sup> /V s)	t (μm)	$ ho$ (m $\Omega$ cm)	$TD (cm^{-2})$
A	$\sim\! 4\times 10^{18}$	107	300 ± 50	10.9 ± 1.8	$1-2 \times 10^{7}$
В	$\sim \! 3 \times 10^{18}$	211	290	7.2	$2-5 \times 10^{5}$
С	$\sim 10^{19}$	150	566	2.7	$< 5 \times 10^{4}$







## Effect of Dislocations on Vertical Diodes & Transistors



Effect of doping on emitter-collector leakage

Figure 5.9: Emitter-Collector leakage is reduced by the addition of a p+ spike in the neutral base of a transistor structure. The solid line represents the emitter-collector leakage of the device with the p<sup>+</sup> blocking layer, while the dashed line is corresponds to the device with a conventional base doped  $p=2 \times 10^{17}$  cm<sup>-3</sup>.

#### Lee McCarthy (PhD Thesis, 2001)

3D Illustration of a band diagram with a dislocation



#### Figure 5.10:

**Above:** Three dimensional rendering of the proposed band diagram of an HBT with a dislocation (arrow) causing local compensation of the p-type base.

**Below:** Calculated cross sectional band-diagram of a locally compensated area surrounding a dislocation in p-type GaN taken at the base-emitter interface. A lightly doped base (left) is fully compensated near the dislocation, while a heavily doped base (right) is only partially compensated. The notches in the cross sectional band diagram correspond to the notch around the dislocation in the three dimensional illustration.



#### Effect of Dislocations on Photonic Devices



FIG. 13 (color). Dependence of LED efficiency on dislocation density for various semiconductor materials. From Lester *et al.*, 1995, Chichibu *et al.*, 1996, and Nakamura, 1998.



### Effect of Dislocations on Photonic Devices

IOP PUBLISHING

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#### Advances in group III-nitride-based deep UV light-emitting diode technology

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**Figure 2.** Simulated IQE versus the dislocation density for 280 nm UV LEDs [17].



## Moving from "GaN" to AIN for electronics



Breakdown Voltage(V)

	Si	GaAs	4H-SiC	GaN	Diamond	β-Ga <sub>2</sub> O <sub>3</sub>	AlN
Bandgap Eg (eV)	1.1	1.4	3.3	3.4	5.5	4.8-4.9	6
Electron mobility $\mu$ (cm <sup>2</sup> /Vs)	1400	8000	1000	1200	2000	300	500
Breakdown field $E_b$ (MV/cm)	0.3	0.4	2.5	3.3	10	8	15
Relative dielectric constant $\epsilon$	11.8	12.9	9.7	9.0	5.5	10	8.5
Baliga's FOM $\epsilon \mu E_b^3$	1	15	340	870	24664	3444	32158



## HEMTs based on the AIN platform



conventional HEMTs



## HEMTs based on the AIN platform





## HEMTs based on the AIN platform





## AIN nucleation on 6H SiC by MBE



Surface treatment: Solvent clean, BHF dip, Ga-Flash in MBE Nucleation: N-Rich AIN (or MEE) at 750 C



## AIN nucleation on AIN/Sapphire by MBE



Surface treatment: Solvent clean Nucleation: N-Rich AIN (or MEE) at 750 C



## AIN nucleation on Bulk AIN substrate by MBE



Surface treatment: Solvent clean Nucleation: N-Rich AIN (or MEE) at 750 C



## AIN nucleation on Bulk AIN substrate by MBE



Nucleation on Bulk AIN Substrates is <u>extremely</u> sensitive to surface contamination. In-situ cleaning requires high temperatures.



# Growth conditions for sharp heterojunctions





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## MBE growth of strained GaN Q-Well HEMTs

SiC QW HEMT on





## MBE growth of strained GaN Q-Well HEMTs



 Top heterojunction is <u>MUCH</u> sharper than the bottom heterojunction

> Increasing GaN Quantum Well thickness

TEM courtesy: Moon Kim (UT Dallas)



## Evolution of strain with GaN Q-Well thickness





## Evolution of strain with GaN Q-Well thickness



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# Strain distribution in Regrown contact HEMTs



## First strained GaN Quantum-Well HEMTs on bulk AIN



AIN/GaN/GaN heterostructure FETs.

and processed sample (top), and a bulk AIN substrate (down).

device.

- We transferred everything we learnt from growth and processing on AIN-on-sapphire templates
- First demonstration of strained GaN QWs on bulk AIN substrates with MBE regrown contacts
- Latest 2DEG mobilities are up to ~600 cm<sup>2</sup>/Vs, but devices fabricated on ~200 cm<sup>2</sup>/Vs samples



## RF results of strained GaN QW HEMTs on bulk AIN



RF performance w/o passivation:

(a) Transfer characteristics of the FETs with 65-nm-long gates in linear scale. (b) Pulsed *I*–*V* measurements with a 500-ns pulse width and a 0.5-ms period. (c) Small-signal RF characteristics of the device showing current gain and unilateral gain  $f_T/f_{max} = 120/24$  GHz.

- Strained GaN QW HEMTs on bulk AIN exhibit current drives ~2.8 A/mm
- RF cutoff frequencies of 120 GHz on ~1.9 A/mm 65 nm gate devices
- Device characteristics beat those on sapphire significantly



# GaN pHEMTs (Gen $I \rightarrow II \rightarrow III \rightarrow IV$ )





## Thermal Conductivity of Compound Semiconductors

$$\begin{aligned} \theta_a^2 &= \frac{5\hbar^2}{3k_B^2} \frac{\int\limits_{0}^{\omega_D} \omega^2 g(\omega) d\omega}{\int\limits_{0}^{\omega_D} g(\omega) d\omega} \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\hbar}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2} dx \\ \kappa_L &= \frac{k_B}{2\pi^2 v} \left(\frac{k_B T}{\tau_C^{-1} (e^x - 1)^2}\right)^3 \int\limits_{0}^{\theta_D/T} \frac{x^4 e^x}{\tau_C^{-1} (e^x - 1)^2$$

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#### Thermal Conductivity of Compound Semiconductors

**Table 2.2.** Calculated and experimental room-temperature thermal conductivity of several zincblende and diamond structure (n = 2) compounds.  $\theta_a =$ high-temperature Debye temperature of the acoustic phonon branch;  $\gamma =$  hightemperature Grüneisen constant;  $\delta^3 =$  volume per atom; M = average atomic mass;  $\kappa_{calc} =$  calculated thermal conductivity from Eq. (2.13);  $\kappa_{exp} =$  measured thermal conductivity.

$\operatorname{Element}/$	$ heta_{m{a}}$	$\gamma$	$\delta$	Μ	$\kappa_{ m calc}$	$\kappa_{ ext{exp}}$
Compound	(K)		(Å)	(amu)	$(\mathrm{Wcm}^{-1}\mathrm{K}^{-1})$	$(\mathrm{Wcm}^{-1}\mathrm{K}^{-1})$
С	1450	0.75	1.78	12.01	16.4	30
Si	395	1.06	2.71	28.08	1.71	1.66
Ge	235	1.06	2.82	72.59	0.97	0.65
BN	1200	0.7	1.81	12.41	11.05	7.6
BP	670	0.75	2.27	20.89	3.59	3.5
$\mathbf{BAs}$	404	0.75	2.39	42.87	1.70	
AlP	381	0.75	2.73	28.98	1.10	
AlAs	270	0.66	2.83	50.95	0.89	0.98
AlSb	210	0.6	3.07	74.37	0.77	0.56
GaP	275	0.75	2.73	50.35	0.72	1.00
GaAs	220	0.75	2.83	72.32	0.55	0.45
GaSb	165	0.75	3.05	95.73	0.33	0.4
InP	220	0.6	2.94	72.90	0.83	0.93
InAs	165	0.57	3.03	94.87	0.51	0.3
InSb	135	0.56	3.24	118.29	0.38	0.2
ZnS	230	0.75	2.71	48.72	0.40	0.27
ZnSe	190	0.75	2.84	72.17	0.35	0.19
ZnTe	155	0.97	3.05	96.49	0.17	0.18
CdSe	130	0.6	3.06	95.68	0.23	
CdTe	120	0.52	3.23	120.00	0.296	0.075



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(Slack)

#### Thermal Conductivity of Compound Semiconductors

**Table 2.3.** Calculated and experimental room-temperature thermal conductivity of several wurtzite (n = 4) compounds.  $\theta_a$  = high-temperature Debye temperature of the acoustic phonon branch;  $\gamma$  = high-temperature Grüneisen constant;  $\delta^3$  = volume per atom; M = average atomic mass;  $\kappa_{calc}$  = calculated thermal conductivity from equation (2.13);  $\kappa_{exp}$  = measured thermal conductivity.

Compound	$egin{aligned}  heta_a\ ({ m K}) \end{aligned}$	$\gamma$	$\delta$ (Å)	M (amu)	$rac{\kappa_{ ext{calc}}}{( ext{Wcm}^{-1} ext{K}^{-1})}$	$rac{\kappa_{ ext{exp}}}{( ext{Wcm}^{-1} ext{K}^{-1})}$
SiC	740	0.75	2.18	20.0	4.45	4.9
AlN	620	0.7	2.18	20.49	3.03	3.5
$\operatorname{GaN}$	390	0.7	2.25	41.87	1.59	2.1
ZnO	303	0.75	2.29	40.69	0.65	0.6
BeO	809	1.38/0.75	1.90	12.51	0.90/3.17	3.7
CdS	135	$0.75^{'}$	2.92	72.23	0.13	0.16



(Slack)

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### High Thermal Conductivity Semiconductors



Fig. 2.7. Thermal conductivity of various single crystals of SiC. R66: pure crystal Slack [65]; #1 and #2: crystals [67] with electron concentrations of  $3.5 \times 10^{16}$  and  $2.9 \times 10^{18}$  cm<sup>-3</sup>, respectively.

Fig. 2.8. Thermal conductivity of GaN as a function of temperature. Open points data of Slack et al. [74]; lower dotted line: results of Sichel and Pankove [71]. Dashe line shows low-temperature boundary limit, assuming a crystal dimension of 50 microns.

Fig. 2.9. Thermal conductivity of single-crystal AlN. "Pure AlN" is the calculated result for a crystal containing no impurities; samples W-201, R-162, and B-21 are single crystals with varying amounts of oxygen concentrations; see text. A ceramic sample is shown for comparison.

Electrically active defects have stronger effect on thermal conductivity than neutral impurities Example: Oxygen significantly degrades the thermal conductivity of AIN, because it creates Aluminum vacancies, that have large dM/M! (Slack)



### Thermal Conductivity of Layered Semiconductors

**Table 2.6.** Thermal conductivity of graphite and BN in the basal plane (xy) and perpendicular to the *c*-axis (z).

Compound	n	$ heta_a$ (K)	$\gamma$	δ (Å)	M (amu)	$rac{\kappa_{ m calc}}{ m (Wcm^{-1}K^{-1})}$	) $(\mathrm{Wcm}^{-1}\mathrm{K}^{-1})$
Graphite-xy	4	1562	0.75	2.05	12.01	27	10-20
Graphite- $z$	4	818	<b>2</b>	2.05	12.01	0.5	0.06
BN-xy	4	1442	0.75	2.05	12.40	22	2 - 3
BN-z	4	755	2	2.05	12.4	0.4	$\sim 0.02$



Fig. 2.6. Thermal conductivity of highly oriented graphite parallel (upper curve) and perpendicular (lower curve) to the basal plane.

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(Slack)

**Table 2.7.** Percentage Increase in Room-Temperature Thermal Conductivity due to the Isotope Effect in Some Group IV and Group III–V semiconductors.

Element/	$\Delta\kappa/\kappa^*$	$\Delta\kappa/\kappa^{**}$	$\Delta \kappa / \kappa^{***}$
Compound	(%)	(%)	(%)
Ge	30	28	30
Si		12	60
С		23	35 - 45
SiC		36	
GaN		5	
BN		125	

\* Model [59]. \*\* Model [60]. \*\*\* Experimental results.



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## Compound Semiconductor Crystal Growth





*Figure 11.2:* Open source evaporation boats (lower) and wire coils (upper figure). Wire coils are generally made of W although other low vapor pressure systems also work. Open boats are made in a wide variety of shapes and of a wide variety of materials including Ta, W, graphite, and other materials.



*Figure 11.1:* Typical evaporation system geometries. Both schematics are simplified and leave out typical components. However, the major parts are indicated. The major difference for the molecular beam epitaxy system is the increased emphasis on ultimate vacuum performance, often four to five orders of magnitude better.

*Figure 11.3:* A simplified schematic sketch of a typical effusion cell. Omitted are the top cap that protects the heater windings from the evaporant as well as improving the black body environment of the crucible, the support clamps and details of the power feedthroughs, etc. The thermocouple junction is pressed gently against the base of the crucible but the black body environment assures an accurate temperature measurement. Note that the heater coils are more closely spaced near the front of the crucible to compensate for greater heat loss in that portion of the cell.



### **Compound Semiconductor Growth**



Fig. 3. Effusion cell used for Cs ion beam experiment for ion propulsion in 1964.

Molecular Beam Epitaxy: Effusion cell design comes from space propulsion systems! Ion beam sources were developed for altitude control, station keeping, and changing orbit.



## Compound Semiconductor Growth



Fig. 4. Effusion cell used for molecular beam epitaxy (photograph courtesy of EPI).



#### Molecular Beam Sources



*Figure 11.4:* The flux distribution of atoms emitted by an effusion cell as a function of the depth in the cell of the evaporant surface relative to the diameter of the surface. [1]. Reprinted from Prog. Crystal Growth and Characterization, Vol 2, Luscher, P.E. and Collins, D.M.; "Design considerations for molecular beam epitaxy systems", pp 15-32, Copyright 1979, with permission from Elsevier.



*Figure 11.6:* Cracking effusion sources for either a solid/liquid source (top) and for a gas source (bottom). The only difference is how the vapor to be cracked is supplied to the baffle section. Normal crackers typically have three baffles.

*Figure 11.5:* The flux patterns resulting from tapered effusion cells in both polar plots (center) and in flux contour plots (right). An isometric schematic shows the design of the tapered effusion cell with a flanged top to reduce creep of effusant out of the cell. Figure redrawn with permission from portions of Figure 3 in Tatuya Yamashita, Takashi Tomita, and Takeshi Sakurai, Japanese Journal of Applied Physics vol. 26 (1987), pages 1192-3. Copywrite 1987, Institute of Pure and Applied Physics.

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#### **Compound Semiconductor Growth**



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#### Materials Sources and Detectors



*Figure 11.7:* A schematic cross section of a typical electron beam evaporator. The magnetic field poles and consequently the magnetic field is out of the plane of the drawing. The high negative potential of the filament assists in electron emission and establishes the electron energy and point of impact on the crucible. Generally electrostatic deflection plates are added to sweep the electron beam into and out of the plane of the drawing.



Figure 11.8: Five methods for monitoring concentrations of gas species. Many others exist.

(Rockett, Chp 11)

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## Tracking crystal growth in-situ using RHEED



*Figure 11.9:* A schematic of the basic apparatus for conducting a RHEED experiment. Not shown is the remainder of the deposition system. This apparatus relies on at least high vacuum to protect the electron gun.

(Rockett, Chp 11)





## Tracking crystal growth in-situ using RHEED



*Figure 11.10:* A schematic of the diffraction patterns in a RHEED experiment resulting from different surface structures. To simplify the drawing the broad rods of the rough surface are drawn as constant in width. Properly, for a two-level surface they have two non-zero Fourier coefficients and should oscillate as sine waves perpendicular to the surface. Therefore the rods should be wider and narrower along their length. See Figure 11.11 for a more accurate representation.

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*Figure 11.11:* Examples of RHEED patterns obtained during non-equilibrium MBE growth of GeSn alloys. (a) Pattern for a very smooth surface as in left side of Figure 11.10 showing points on an arc. The diagonal streaks are multiple diffraction effects known as Kikuchi bands. (b) A pattern for a somewhat rougher surface corresponding to the middle drawing in Figure 11.10. Both (a) and (b) show a half-order reflection between the bright integer-order spots that results from the surface reconstruction. (b) Shows some evidence of residual Kikuchi bands. (c) A pattern for a very rough surface corresponding to the right hand drawing in Figure 11.10 where the electron beam is passing through three dimensional islands. Adapted with permission from O. Gurdal, P. Desjardins, J. R. A. Carlsson, N. Taylor, H. H. Radamson, J.-E. Sundgren, and J. E. Greene, Journal of Applied Physics, 83, 162 (1998). Copyright 1998, American Institute of Physics.



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## Tracking crystal growth in-situ using RHEED



Figure 11.12: RHEED diffraction Ewald sphere constructions for the on-Bragg and off-Bragg of conditions. Note that in the on-Bragg condition there is no difference in the diffracted intensity for the rough and smooth surfaces while for the off-Bragg condition alternate layers of the film interfere destructively. Therefore there is strong contrast for the intensity from rough and smooth surfaces. Refer to Figure 4.6 and related discussion for background on this construction.



Figure 11.13: A typical example of (hypothetical) RHEED oscillation data during MBE growth at a rate of 1 monolayer/sec. Note that the surface is roughening during the 8 monolayers of deposition as shown by the decay of the oscillations but upon termination of growth the surface becomes smooth and the intensity recovers. To obtain this type of behavior the experiment is configured in the off-Bragg condition so that alternate monolayers of the film interfere destructively. This type of oscillation is observed in the intensity of all reflected and diffracted beams. The most intense is the specularly-reflected beam so that is the one normally used.



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Figure 6.1: Terraces, ledges, kinks, adatoms, adatom clusters, advacancies, and advacancy clusters on a Kossel crystal.

(Rockett chp 10,11, Tsao)

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Figure 6.2: x-T phase diagrams for surface roughening. Above and below each phase diagram are also shown the normalized molar free energies of the adlayer phases at kT/w = 2.5 and kT/w = 1.5, their common tangents, and the critical compositions (open circles) determined by those common tangents.

 $T_{\rm r,adat} \approx \frac{2}{3} \Delta h_{
m sub}$ 

(Rockett chp 10,11, Tsao)

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Figure 6.3: A step wandering on the surface of a Kossel crystal due to randomly distributed kinks.



Figure 6.4: In moving an atom from position a to position b, two net bonds are broken, but four kinks are formed. If the bond energy is w, then the energy per kink is  $\epsilon_{\text{kink}} = 2w/4 = w/2$ .

$$f_{\text{step}} = \epsilon_{\text{step}} + \epsilon_{\text{kink}}(p_{+} + p_{-}) + kT[(p_{\text{ext}} + p_{\text{int}}/2) \ln(p_{\text{ext}} + p_{\text{int}}/2) + (p_{\text{int}}/2) \ln(p_{\text{int}}/2) + (1 - p_{\text{ext}} - p_{\text{int}}) \ln(1 - p_{\text{ext}} - p_{\text{int}})].$$
(6.9)

(Rockett chp 10,11, Tsao)

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$$f_{\text{step}} = \epsilon_{\text{step}} + \epsilon_{\text{kink}}(p_{+} + p_{-}) \\ + kT[(p_{\text{ext}} + p_{\text{int}}/2) \ln(p_{\text{ext}} + p_{\text{int}}/2) + (p_{\text{int}}/2) \ln(p_{\text{int}}/2) \\ + (1 - p_{\text{ext}} - p_{\text{int}}) \ln(1 - p_{\text{ext}} - p_{\text{int}})].$$

$$\frac{\partial f_{\text{step}}}{\partial r} = \epsilon_{\text{kink}} + kT[(1/2) \ln(p_{\text{ext}} + p_{\text{int}}/2) + (1/2) \ln(p_{\text{int}}/2)]$$

$$\partial p_{\text{int}}$$
  
 $-(1/2)\ln(1 - p_{\text{ext}} - p_{\text{int}})]$   
 $= 0.$ 

$$p_+p_-=p_o^2e^{-2\epsilon_{\rm kink}/kT}$$

$$p_{o}^{\text{equ}} = \frac{1 - \sqrt{1 - (1 - 4e^{-2\epsilon_{\text{kink}}/kT})(1 - \tan^{2}\phi)}}{1 - 4e^{-\epsilon_{\text{kink}}/kT}}$$

$$p_{+}^{\text{equ}} = \frac{\tan\phi}{2} + \sqrt{p_{o}^{2}e^{-2\epsilon_{\text{kink}}/kT} + \frac{1}{4}\tan^{2}\phi}}{p_{-}^{\text{equ}}} = -\frac{\tan\phi}{2} + \sqrt{p_{o}^{2}e^{-2\epsilon_{\text{kink}}/kT} + \frac{1}{4}\tan^{2}\phi}}.$$



Figure 6.6: Dependences of the energy, entropy, and free energy of a step on the probability of intrinsic kinks. The step is assumed to have no extrinsic kinks, and a kink is assumed to add an additional energy equal to half the energy of a straight step. The energies and free energies are normalized to the energy of a straight step. As the temperature increases, the intrinsic kink probability that minimizes the free energy of the step increases, and the minimum free energy decreases.

(Rockett chp 10,11, Tsao)

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#### Growth Physics: Connection to Ising Models

**Exercise:** In a lattice gas model for crystal growth, a density variable  $n_i$  at a lattice site *i* takes a value 1 when the site is occupied by an atom, or it is 0 when the site is empty. If a broken chemical bond from a crystal atom costs an energy J, the interaction energy is expressed as

$$\mathcal{H} = J \sum_{\langle ij \rangle} [n_i(1 - n_j) + (1 - n_i)n_j] - \Delta \mu \sum_i n_i, \qquad (8.1)$$

where  $\langle ij \rangle$  means the summation over the nearest neighbor sites pairs, and  $\Delta \mu$  is the chemical potential gain by crystallization. Show that the Hamiltonian (8.1) can be transformed to the two-dimensional Ising Hamiltonian

$$\mathcal{H}_{\mathrm{I}} = -\frac{J}{2} \sum_{\langle ij \rangle} S_i S_j - H \sum_i S_i. \tag{8.2}$$

Here an Ising spin variable  $S_i = 2n_i - 1$  takes values +1 or -1, and the field is  $H = \Delta \mu/2$  [128].



$$\frac{\beta(\theta)}{k_B T} = |\cos\theta|\sinh^{-1}(\alpha|\cos\theta|) + |\sin\theta|\sinh^{-1}(\alpha|\sin\theta|)$$

$$\alpha = \frac{2}{b} \left( \frac{1 - b^2}{1 + \sqrt{\sin^2 2\theta} + b^2 \cos^2 2\theta} \right)^{1/2}$$
$$b = \frac{2 \sinh(J/k_B T)}{\cosh^2(J/k_B T)}.$$

(Saito, Stat. Phy. of Crystal Growth)

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 $\frac{k_B T_c}{J} = \frac{1}{\ln(\sqrt{2}+1)} = 1.1346\cdots$ 

 $\mathcal{H} = -\frac{J}{2} \sum_{\langle ij \rangle} S_i S_j + \frac{NzJ}{4}$ 

$$\begin{split} f_{\text{surf}} &= f_{\text{terr}} + f_{\text{step}} \tan \theta \\ b^2 &\equiv p_+ a^2 + p_- a^2 \\ \frac{\Delta s_{\text{step}}}{k} &= \frac{\pi^2}{6} \frac{b^2/2}{l^2} = \frac{\pi^2}{12} \frac{b^2}{a^2} \tan^2 \phi \\ f_{\text{surf}} &= f_{\text{terr}} + f_{\text{step}} \tan \phi + kT \frac{\pi^2}{12} \frac{b^2}{a^2} \tan^3 \phi \\ \Delta u_{\text{step}} &= A \left[ \frac{1}{(l+x)^2} + \frac{1}{(l-x)^2} \right] \approx \frac{2A}{l^2} (1 + x^2/l^2) \end{split}$$



Figure 6.7: Five wandering steps of average spacing l. Two of the steps intersect, by chance, three times.

$$\frac{d(\Delta u_{\text{step}} - T\Delta s_{\text{step}})}{dx} = \frac{d}{dx} \left[ \frac{2A}{l^2} \left( 1 + \frac{x^2}{l^2} \right) + kT \frac{\pi^2}{12} \frac{b^2}{x^2} \right] = 0$$
$$d = \left( \frac{\pi^2}{24} \frac{kTb^2}{A} \right)^{1/4} l$$

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$$\begin{split} f_{\text{surf}} &= f_{\text{terr}} + f_{\text{step}} \tan \theta, \\ b^2 &\equiv p_+ a^2 + p_- a^2 \\ \frac{\Delta s_{\text{step}}}{k} &= \frac{\pi^2}{6} \frac{b^2/2}{l^2} = \frac{\pi^2}{12} \frac{b^2}{a^2} \tan^2 \phi \\ f_{\text{surf}} &= f_{\text{terr}} + f_{\text{step}} \tan \phi + kT \frac{\pi^2}{12} \frac{b^2}{a^2} \tan^3 \phi \\ \Delta u_{\text{step}} &= A \left[ \frac{1}{(l+x)^2} + \frac{1}{(l-x)^2} \right] \approx \frac{2A}{l^2} \left( 1 + \frac{x^2}{l^2} \right) \end{split}$$

$$\frac{d(\Delta u_{\text{step}} - T\Delta s_{\text{step}})}{dx} = \frac{d}{dx} \left[ \frac{2A}{l^2} \left( 1 + \frac{x^2}{l^2} \right) + kT \frac{\pi^2}{12} \frac{b^2}{x^2} \right] = 0$$
$$d = \left( \frac{\pi^2}{24} \frac{kTb^2}{A} \right)^{1/4} l.$$

$$\Delta f_{\text{step}} = \frac{2A}{l^2} \left( 1 + \sqrt{\frac{\pi^2 k T b^2}{6A}} \right)$$
  

$$f_{\text{surf}} = f_{\text{terr}} + (f_{\text{step}} + \Delta f_{\text{step}}) \tan \theta$$
  

$$= f_{\text{terr}} + f_{\text{step}} \tan \theta + \frac{2A}{a^2} \left( 1 + \sqrt{\frac{\pi^2 k T b^2}{6A}} \right) \tan^3 \theta.$$



Figure 6.8: Changes in energy, entropy, and free energy as the width of the alley (x/a) in units of lattice constants) within which a step is allowed to wander increases. Both the energy and entropy increase as the width of the alley increases and approaches the mean step spacing (l/a) in units of lattice constants). The equilibrium alley width (d/a) in units of lattice constants) is that which minimizes the free energy. As the ratio between the temperature and the strength of the interaction between steps (kT/A) increases, both the equilibrium alley width and the free energy at that equilibrium alley width increase.

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Figure 6.5: Scanning tunneling micrograph of a Si surface misoriented  $0.5^{\circ}$  from (001) toward [110]. The surface height decreases from upper left to lower right.<sup>*a*</sup> On this surface, alternate single-height steps are referred to as type *SA* and *SB*, and are smooth and rough, respectively, reflecting the relative energies of kink formation.

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(Rockett chp 10,11, Tsao)





$$E = 2\gamma_x h_y h_z + 2\gamma_y h_x h_z + 2\gamma_z h_x h_y$$

$$E = \frac{\gamma_x V}{4h_x} + \frac{\gamma_y V}{4h_y} + 2\gamma_z h_x h_y$$

$$\gamma_x h_y h_z = \gamma_y h_x h_z = \gamma_z h_x h_y = (\gamma_x \gamma_y \gamma_z)^{1/3} V^{2/3} = \text{constant}$$

$$\frac{\gamma_x}{h_x} = \frac{\gamma_y}{h_y} = \frac{\gamma_z}{h_z} = \left(\frac{\gamma_x \gamma_y \gamma_z}{V}\right)^{1/3}$$

(Rockett chp 10,11, Tsao)

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