Journal of Crystal Growth 482 (2018) 61-69



Contents lists available at ScienceDirect

Journal of Crystal Growth

journal homepage: www.elsevier.com/locate/crys

MBE growth of few-layer 2H-MoTe₂ on 3D substrates

Suresh Vishwanath ^{a,b,*}, Aditya Sundar ^c, Xinyu Liu ^d, Angelica Azcatl ^e, Edward Lochocki ^f, Arthur R. Woll ^g, Sergei Rouvimov ^b, Wan Sik Hwang ^b, Ning Lu ^e, Xin Peng ^e, Huai-Hsun Lien ^c, John Weisenberger ^b, Stephen McDonnell ^{e,h}, Moon J. Kim ^e, Margaret Dobrowolska ^d, Jacek K. Furdyna ^d, Kyle Shen ^{f,i}, Robert M. Wallace ^e, Debdeep Jena ^{a,b,c}, Huili Grace Xing ^{a,b,c,i}

^a School of Electrical and Computer Engineering, Cornell University, Ithaca, NY 14853, United States

^b Department of Electrical Engineering, University of Notre Dame, IN 46556, United States

^c Department of Materials Science and Engineering, Cornell University, Ithaca, NY 14853, United States

^d Department of Physics, University of Notre Dame, Notre Dame, IN 46556, United States

^e Department of Materials Science and Engineering, University of Texas Dallas, Dallas, TX 75083, United States

^f Department of Physics, Laboratory of Atomic and Solid State Physics, Cornell University, Ithaca, NY 14853, United States

^g Cornell High Energy Synchrotron Source, Cornell University, Ithaca, NY 14853, United States

^h Department of Materials Science and Engineering, University of Virginia, Charlottesville, VA 22904, United States

ⁱ Kavli Institute at Cornell for Nanoscale Science, Cornell University, Ithaca, NY 14853, United States

ARTICLE INFO

Article history: Received 10 July 2017 Received in revised form 2 October 2017 Accepted 19 October 2017 Available online 20 October 2017 Communicated by A. Bhattacharya

Keywords: A3. Molecular beam epitaxy B1. Tellurites B2. Semiconducting materials A1. X-ray diffraction

ABSTRACT

MoTe₂ is the least explored material in the Molybdenum-chalcogen family. Molecular beam epitaxy (MBE) provides a unique opportunity to tackle the small electronegativity difference between Mo and Te while growing layer by layer away from thermodynamic equilibrium. We find that for a few-layer MoTe₂ grown at a moderate rate of ~6 min per monolayer, a narrow window in temperature (above Te cell temperature) and Te:Mo ratio exists, where we can obtain pure phase 2*H*-MoTe₂. This is confirmed using reflection high-energy electron diffraction (RHEED), Raman spectroscopy and X-ray photoemission spectroscopy (XPS). For growth on CaF₂, Grazing incidence X-ray diffraction (GI-XRD) reveals a grain size of ~90 Å and presence of twinned grains. In this work, we hypothesis the presence of excess Te incorporation in MBE grown few layer 2*H*-MoTe₂. For film on CaF₂, it is based on >2 Te:Mo stoichiometry using XPS as well as 'a' and 'c' lattice spacing greater than bulk 2*H*-MoTe₂. On GaAs, its based on observations of Te crystallite formation on film surface, 2 × 2 superstructure observed in RHEED and low energy electron diffraction, larger than bulk c-lattice spacing as well as the lack of electrical conductivity modulation by field effect. Finally, thermal stability and air sensitivity of MBE 2H-MoTe₂ is investigated by temperature dependent XRD and XPS, respectively.

© 2017 Elsevier B.V. All rights reserved.

癥

CRYSTAL

CrossMark

1. Introduction

MoTe₂ is still a relatively unexplored transitional metal dichalcogenide (TMD) and holds great promise. MoTe₂ exists in trigonal prismatic (2*H* structure) semiconducting phase at room temperature and metallic 1*T* phase (monoclinic structure) at high temperatures or metallic T_d phase (orthorhombic structure) when metastable 1*T* is cooled to $-33 \degree$ C to $-13 \degree$ C [1]. 1*T* and T_d crystals have the same in-plane crystal structures but vary in vertical stacking. Monolayer 2*H*-MoTe₂ is predicted to be among the smallest

bandgap semiconducting TMDs [2]. Using scanning tunneling spectroscopy (STS) measurements, the bandgap of monolayer 2*H*-MoTe₂ has been measured to be between 1.03 eV [3] to 1.4 eV [4]. 2*H*-MoTe₂ has a close to broken band alignment with materials like SnSe₂ [5], making them attractive for tunnel based devices such as Esaki diodes [6] and two-dimensional heterojunction interlayer tunneling field effect transistors (Thin-TFETs) [7,8]. The T_d -MoTe₂ is a type II Weyl semimetal [1], which is a new class of topological material. Traditionally, the transition between 2*H* and 1*T* phase was thought to be abrupt with respect to temperature [9] but recently a mixed phase region has been observed in the phase diagram [10]. Even for growth under UHV conditions, coexistence of 2*H* and 1*T* phase has been recently reported [11]. The transition between the 2*H* and 1*T* phase holds promise for applications such as low resistance contacts [12] and phase change memory.

^{*} Corresponding author at: School of Electrical and Computer Engineering, Cornell University, Ithaca, NY 14853, United States.

E-mail addresses: sv372@cornell.edu (S. Vishwanath), grace.xing@cornell.edu (H.G. Xing).

Large area, phase-pure epitaxial growth with layer controllability would enable industrial applications, but growth of MoTe₂ is especially challenging, since the electronegativity difference between Mo and Te is just 0.3 eV [13–15], resulting in a weak bond. Until recently, MoTe₂ has been obtained using tellurization of e-beam deposited molybdenum or molybdenum oxide films [16–18] or by chemical vapor transport (CVT) [19,10]. In case of tellurizing Mo, 1*T*'-MoTe₂ phase is initially formed at 650 °C, which can be converted to 2H-MoTe₂ by a 3 h anneal under Te vapor [17]. Tellurizing MoO₃ transforms to 2H-MoTe₂ more readily, but one cannot be sure in achieving 100% percent reduction of oxygen. For CVT grown MoTe₂ a mixed phase is observed in the growth temperature range of 500–900 °C based on the tellurium content in MoTe_x [10], but under ultrahigh vacuum (UHV) in nonthermodynamic equilibrium conditions, this phase boundary is unknown. For example, the formation of a new metallic nanowire phase has been recently reported upon Te loss by annealing at 400-500C in UHV [20]. We recently demonstrated growth of 2H-MoTe₂ using molecular beam epitaxy (MBE) in a superlattice with MoSe₂ as well as Bi_2Te_3 [21], where we used a growth temperature of 380 °C. Growth temperature of ~340 °C [22] or 200 °C [3,4] have been used in other recent MBE demonstrations of 2H-MoTe₂ growth. Also, work on phase purity and stability for MBE 2H-MoTe₂ has been done only on monolayer MoTe₂ grown on layered substrates [3,4,11].

In this study, we chose 340 °C to be the lower bound of growth temperature, so as to keep the growth temperature higher than the Te cell temperature (~300 °C). This avoids intentional accumulation of Te, while keeping the growth temperature significantly lower than the lower bound (500 °C) of the mixed phase, as suggested by the phase diagram under 1 ATM of Te vapor [10]. We observe that even at a substrate growth temperature as low as 340 °C, the crystalline phase of the MBE-grown MoTe₂ has a sensitive dependence on uncracked Te flux in a Te rich environment (Te: Mo > 70 for all growths). This is presented by analyzing a series of 3 samples of MoTe₂ under different growth conditions on CaF₂ and identifying a growth condition for growth of pure phase 2H-MoTe₂. Here, we show that it is indeed feasible to grow phase-pure 2H-MoTe₂ on a CaF₂ substrate without any requirement of a postgrowth anneal. We also present the growth and electrical characterization of 2H-MoTe₂ on GaAs (111) B. The transition to GaAs was motivated by the availability of high quality epi-ready n⁺ GaAs substrates necessary for characterization using techniques such as low energy electron diffraction (LEED). Table 1 summarizes the growth conditions of all the 5 samples in this study. The Mo flux is calculated using the experimentally determined growth rates while assuming zero desorption for molybdenum adatoms, which enables calculation of the Te:Mo flux ratio tabulated in Table 1 for all samples. Growth rate on CaF₂ is calculated using crosssection transmission electron microscopy image and on GaAs

Table 1							
Growth	conditions	for	all	samples	in	this	study.

. . . .

(111) B from Reflection high-energy electron diffraction (RHEED) oscillations.

2. Growth conditions

2.1. Growths on CaF₂ substrate

Three samples (see Table 1) constitute the series of samples grown on CaF₂. Calcium fluoride was chosen as a substrate because (i) it has an inert fluorine-terminated surface on which we have successfully grown MoSe₂ [23] and (ii) it provides a cavity effect enhancing the Raman signal due to its optical transparency. On the other hand, the Raman signal from MoTe₂ grown on GaAs (111) B is very weak [21]. These samples were grown in a Riber 32 MBE system using elemental Mo delivered from an e-beam evaporator and elemental uncracked Te from a Knudsen cell. The growth duration was 30 min for each sample and the Mo flux, which limits the growth rate, was set to ~ 0.17 monolayer (ML) per minute or 6 min per ML. Supplementary (SI) Fig. 1(a) shows that all CaF_2 substrates were first heated to 800 $^\circ\text{C},$ held for 30 min in order to anneal and degas. Sharp RHEED streaks of CaF₂ prior to start of growth (see Fig. 1(a)), show the smooth crystalline post-anneal growth surface. Then the substrates were lowered to respective growth temperatures, stabilized for \sim 30 min prior to thin film growth. The growth conditions are listed in Table 1. Using these 3 samples, we observe that, although the Te flux is \sim 2 orders of magnitude higher than the Mo flux, the Te flux range to obtain 2H-MoTe₂ is narrow and the substrate temperature control is critical. None of the MoTe₂ films on CaF₂ have been annealed in order to avoid phase change during annealing. All temperatures given in this study are thermocouple temperatures and the sample surface temperature is estimated about 20 °C lower than the thermocouple temperature.

All samples on CaF_2 where grown on the same day under identically Mo flux conditions. Hence, identical film thickness of ~5ML is expected for the three samples on CaF_2 . TEM was done on sample A and sample B, discussed further below, to confirm similar film thickness.

2.2. Growths on GaAs substrate

Two samples, sample D and sample E, of $MoTe_2$ on GaAs (111)B are discussed. Post-growth anneal was done on $MoTe_2$ films on GaAs.

Sample D was grown at a growth temperature of 340 °C and Te flux of 6.9×10^{-6} Torr (slightly higher than Sample A due to higher thermal conductivity of GaAs than CaF₂). SI Fig. 1(b), shows the growth sequence. The key step in this growth is the anneal of GaAs under Te prior to MoTe₂ growth to achieve smoother Te terminated

Sample ID	Substrate	MoTe ₂ phase	Substrate temperature (°C)	Te flux (Torr)	Te:Mo flux ratio	Growth duration (mins)	Post growth anneal
А	CaF ₂ (111)	2H	340	6.5×10^{-6}	297	30	None
В	CaF ₂ (111)	2H + UP*	340	$\textbf{2.0}\times \textbf{10}^{-6}$	98	30	None
С	CaF ₂ (111)	UP^*	400	1.4×10^{-6}	71	30	None
D	GaAs (111) B	2H	340	$\textbf{6.9}\times 10^{-6}$	175	30	at 380 °C for 10 min without Te
E	GaAs (111) B	2H	340	6.9×10^{-6}	262	20	at 450 °C for 3 min and 550 °C for 7 min under Te

*UP stands for unidentified phase that has some similarity to 1T' but is not identical.

Growths on CaF,



Fig. 1. (a–d) RHEED pattern from the series of samples on CaF_2 showing evolution of polycrystalline growth with lowering Te:Mo flux ratio and increasing growth temperature as compared to well-aligned growth of 2H phase. Sample D: RHEED post growth of 2H-MoTe₂ on GaAs (e) pre anneal and (f) post anneal are shown. The dashed lines are guide to the eye, yellow for MoTe₂ and red for the additional set of lines observed. (g) RHEED oscillations at the spectral point during growth of MoTe₂ shows a close to layer by layer growth with a period of ~218 s per monolayer. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

surface. This is consistent with our previous report of Te anneal of GaAs [21] prior to TMD superlattice growth.

Sample E was grown on n⁺ GaAs for characterizations requiring conducting substrate. Prior to the post-growth anneal, the growth sequence for sample E was identical to sample D. The post-growth anneal for sample E was done at 450 °C for 3 min and 550 °C for 7 min under Tellurium flux. After this post-growth anneal, the sample was cooled under Te till growth temperature of 340 °C and then capped with ~100 nm Se during cool-down to room temperature for surface protection during sample transfer to other characterization tools. The change in growth rate between samples D and E (see Table 1) is due to variability of Mo flux at the same e-beam power on different growth days.

3. Results and discussion

3.1. Reflection high-energy electron diffraction (RHEED)

3.1.1. RHEED on MoTe₂ films on CaF₂ substrate

As seen in Fig. 1(a), annealed CaF₂ has a strong streaky RHEED pattern prior to growth. Sample A shows a more diffused but still streaky RHEED pattern, whereas sample B shows a mixed pattern comprising of streaks and a ring, which evolves into only a ring in sample C. Streaky RHEED points to layered growth with minimal mosaicity of the as-grown film, but progressive inclusion of a ring points to another growth mechanism taking over, which results in polycrystalline growth. Whether the polycrystalline material is the same phase as the streaky film or a different phase is elucidated through employing Raman and XPS characterization (described below). The RHEED streaks of MoTe₂ in sample A along $\langle 11\bar{2}0 \rangle$

appear at the same position as the $\langle 1 \bar{1} 0 \rangle$ of CaF₂, as observed previously in MoSe₂ on CaF₂ [23,24]. The in-plane lattice spacing of MoTe₂ based upon the ratio of the RHEED streak spacing is ~3.5 Å, which is very close to the value of 3.52 Å [10] corresponding to bulk 2*H*-MoTe₂. The ambiguity in this measurement is due to the diffused RHEED pattern, which is measured more accurately using grazing incidence X-ray diffraction (GI-XRD), presented further below.

3.1.2. RHEED on MoTe₂ films on GaAs substrate

For sample D, a pair of faint RHEED streaks with a spacing less than 2H-MoTe₂ were observed. Te has a hexagonal crystal structure with lattice constants of a = 4.456 Å and c = 5.921 Å [25]. Since the inplane lattice constant of Te is greater than that of 2H-MoTe₂, it was the initial suspect. With the aim to remove any excess Te in the film, a post-growth anneal at 380 °C without any Te flux was done. But, as seen in Fig. 1(e and f), the anneal doesn't remove this second set of streaks. Further analysis reveals that the ratio of spacing of the RHEED streaks from MoTe2 and the newly observed streaks is \sim 2. If the lines were from 2H-MoTe₂ and Te, the expected ratio is ~1.2. So, its likely not due to elemental tellurium at the surface. One hypothesis is presence of ordered defects, which could be Te interstitials. Fig. 1(g) shows RHEED intensity oscillations of the RHEED spectral point during the growth of MBE 2H-MoTe₂. This shows close to layer by layer growth. Simulation in SI Fig. 2 shows the crest and the trough do not necessarily indicate a complete monolayer, variation in smoothness can cause shifts. But approximately, the period between crests corresponds to a monolayer. Increasing roughness or waviness in the film is the likely cause for decay with RHEED oscillation intensity in Fig. 1(g).



Fig. 2. (a) Raman measurements on the samples grown on CaF₂. (b) Raman from MoTe₂ grown on GaAs post-exfoliation, using scotch tape, on to SiO₂/Si for a better signal.

Table 2 Positions of the various Raman peaks compared to measured values for bulk 2H-MoTe₂ and reported values for 1T-MoTe₂ [17].

Sample ID	$\begin{array}{c} A_{1g} \ position \\ (cm^{-1}) \end{array}$	E_{2g}^2 position (cm ⁻¹)	B _g position (cm ⁻¹)	A _g position (cm ⁻¹)
Bulk 2 <i>H</i>	174	236	NA	NA
A	173	236	NA	NA
B	174	237	158	256
C	NA	NA	159	256
1 <i>T</i> ′ [17]	NA	NA	163	256.1

3.2. Raman spectroscopy

3.2.1. Raman on MoTe₂ films on CaF₂ substrate

Raman spectra in Fig. 2(a) confirm an evolution from the 2H phase to a new phase as we progress from sample A to sample C. The 2H phase is confirmed by comparing Raman from sample A with Raman from CVT grown bulk 2H-MoTe₂ obtained from 2D Semiconductors Inc. as shown in Table 2. Its important to note that the FWHM of the peaks from MBE grown MoTe₂ is several times wider than that of the CVT-grown MoTe₂. This points to a significant disorder in the MBE-grown material and augments the observation of the diffuse MoTe₂ RHEED pattern in sample A. The new phase is labeled as the unidentified phase (UP) MoTe_x in this work. This is because, as seen in Table 2 the Ag peak position of the UP phase at $\sim 256 \text{ cm}^{-1}$ agrees closely with the reported value of 256.1–257 cm⁻¹ for 1T' MoTe₂ growth by tellurization of molybdenum films but the peak at $158-159 \text{ cm}^{-1}$ deviates significantly from the reported value of $163-161 \text{ cm}^{-1}$ for B_g peak from 1TMoTe₂ and is extremely broad [17]16. Sample B shows a mixed phase comprised of Raman signatures from both 2H and UP phases. Sample C is comprised mostly of UP phase but due to the broad peaks, presence of some 2H phase MoTe₂ cannot be ruled out.

3.2.2. Raman on MoTe₂ films on GaAs substrate

Since, the Raman signal from MoTe₂ on GaAs is quite weak [21], the as grown film was exfoliated and transferred to SiO₂/Si substrate using a scotch tape. The transfer was performed to enhance the raman signal from the MoTe₂ due to cavity effect from SiO₂ as well as to eliminate the interference from the LO phonon raman peaks from GaAs. In Fig. 2(b), the peaks below 150 cm⁻¹ can be attributed to Te [26]. E_{2g} peak from transferred MoTe₂ from sample D is almost symmetric and peak position is consistent with that from bulk MoTe₂ at 235.8 cm⁻¹. The reason for broadening in the A_{1g} peak is unclear.

3.3. X-ray photoemission spectroscopy (XPS)

3.3.1. XPS on MoTe₂ films on CaF₂ substrate

XPS spectra corresponding to Mo, Te, O, Ca, F and C are detected from all samples (Fig. 3). No charging effects were detected on any of them. Peak positions for Mo $3d_{5/2}$ and Te $3d_{5/2}$ as well as Te:Mo ratio corresponding to both phases 2H-MoTe₂ and UP-MoTe_x are listed in Table 3. For sample A, the Mo 3d_{5/2} signal corresponding to 2H-MoTe₂ bond was detected at 227.9 eV, which is consistent with the binding energy of 2H-MoTe₂ in literature [14]. The Te 3d_{5/2} peak corresponding to the 2H-MoTe₂ is observed at 572.8 eV. Molybdenum oxide in the Mo⁺⁶ state was also identified. In the Te 3d spectrum, tellurium dioxide and MoTe₂ are both detected. The Te:Mo ratio is ~2.57 after correction due to attenuation from the oxide overlayer. For sample B, in addition to the peaks corresponding to 2*H*-MoTe₂, Mo⁺⁶ oxide and TeO₂ are detected; the Mo $3d_{5/2}$ peak at 227.6 eV and the Te $3d_{5/2}$ peak at 572.6 eV are assigned to the UP phase. The Te: Mo ratio for the 2H-MoTe₂ component is calculated to be 2.03. After correction due to attenuation from the oxide overlayer the Te:Mo ratio is calculated to be 2.54. The Te:Mo ratio for the UP-MoTe_x component in sample B is 2.12 and, when corrected for the oxide overlayer is 2.66. For sample C, there is a very small signal from 2H-MoTe₂ with Mo 3d_{5/2} at 228 eV and the corresponding peak for Te $3d_{5/2}$ at 572.7 eV. But the majority of the MoTe₂ peak intensity is from a new Mo $3d_{5/2}$ peak at 227.5 eV and Te $3d_{5/2}$ at 572.5 eV, which are assigned to chemical states associated with the UP phase of MoTex. The Te: Mo ratio corresponding to UP phase of $MoTe_2$ is 2.13 and, when corrected for the oxide overlayer is 2.72. The Te:Mo ratio for the 2H-MoTe₂ component is 2.04 and, when corrected for the oxide overlayer, is 2.61. It is key to note that in sample C the Mo 3d peak intensity associated with Mo oxide is much higher than that for MoTe₂, as well as the oxide intensity from the other samples. This suggests that, in spite of employing a large over pressure of uncracked Te (dimers) during growth, not only does Mo form predominantly the UP-MoTe_x but that majority of Mo has an increased propensity for oxidation. This molybdenum oxide in sample C exhibits 2 different Mo oxidation states of +5 and +6. Reported peak position for 1T-MoTe₂ for Mo 3d_{5/2} is 227.7–228 eV and for Te 3d_{5/2} is 572.1–572.6 eV [27,16]. The observed XPS peak from the phase assigned to UP-MoTe_x for Te $3d_{5/2}$ is consistent with the reported value but that for Mo $3d_{5/2}$ is much lower that what has been reported for any Mo-Te bond and even metallic Mo 3d_{5/2} at 227.8 eV [28]. It is also noted that in all 3 samples, the oxide peaks from Mo and Te in the O1s spectra could not be resolved because of the close proximity in electronegativity of Mo and Te [13]. The O 1s spectral feature also has



Fig. 3. (a) XPS on samples showing various phases and the extent of oxidation under different growth conditions. The pink line corresponds to Mo^{+6} oxide, maroon line to 2H-MoTe₂, blue line to UP-MoTe_x (the unidentified phase), red line to TeO₂ and green line to Mo^{+5} oxide. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

XPS peak positions for Mo 3d_{5/2} and Te 3d_{5/2} as well as Te:Mo ratio corresponding to both phases 2*H*-MoTe₂ and *UP*-MoTe_x compared to reported values.

Sample ID	Te:Mo flux ratio	Mo 3d _{5/2} (eV)	Te 3d _{5/2} (eV)	Te:Mo ratio w/o corr.	Te:Mo ratio corr.
Bulk 2H		227.8 (2 <i>H</i>) (air exposed [14])	572.9 (2 <i>H</i>) (air exposed [14])		2.03 (2H)
		228.0 (2H) (as cleaved [14])	573.0 (2H) (as cleaved [14])		
		227.8 (2H) (as cleaved [13])	572.4 (2H) (as cleaved [13])		
А	297	227.9 (2H)	572.8 (2H)		2.57 (2H)
В	98	227.9 (2H)	572.9 (2H)	2.03 (2H)	2.54 (2H)
		227.4 (UP)	572.6 (UP)	2.12 (UP)	2.66 (UP)
С	71	228 (2H)	572.9 (2H)	2.04 (2H)	2.61 (2H)
		227.5 (UP)	572.6 (UP)	2.13 (UP)	2.72 (UP)
1 <i>T</i> ′		227.7 (11') [27]	572.1 (1 <i>T</i> ') [27]		
		228 (17') [16]	572.6 (1 <i>T</i> ') [16]		

contributions from C-O and O-H. SI Fig. 4(a and b) shows that concentration of both molybdenum oxide and tellurium oxide concentrations are higher on the surface as seen from the increase in intensity of oxides at take-off angle of 45° as compared to 80° . (A take-off angle of 80° is much more bulk sensitive than 45° .) Also, the chemical bonding state of MoTe₂ is homogeneous through the analyzed depth, as the Mo-Te peak width remains constant when changing angle. Its worthy to note that the extent of oxidation in the telluride system is much more than previously reported MBE MoSe₂ [23].

Table 3

3.3.2. Capped growth on GaAs substrate to study sensitivity to air exposure of MBE-MoTe₂

To understand the ease of oxidation of MoTe₂, on sample E, the Se cap was removed from Sample E by heating in a UHV system, followed by in situ XPS. After the initial XPS measurements, the sample was exposed to air for 20 min and then XPS done again. It is seen from SI Fig. 4(c-d) that oxide peaks appear in both Mo and Te XPS spectrum. Approximately, 8% of the surface area under goes oxidation in 20 min. The Se decapping is done in an oxide MBE system that is connected to LEED and XPS system to avoid air exposure. The O 1s signal observed in XPS (see SI Fig. 4(e)) prior to air exposure is likely due to physisorbed oxygen, postdecapping, from the oxide MBE chamber (base pressure of $\sim 1 \times$ 10^{-8} Torr). This is consistent with the fact that electronegativity difference between Mo and Te is 0.3 eV but that between O and Te is 1.4 eV, making the compound prone to oxidation [14,13]. Effect of air exposure in MBE grown films could be exacerbated by oxidation at the edges of MoTe₂ grains, similar to that reported for MBE grown WSe₂ films [29].

3.4. X-ray diffraction (XRD) and transmission electron microscopy (TEM)

3.4.1. MoTe₂ films on CaF₂ substrate

In order to get a better estimation of the inplane lattice constant of MBE 2H-MoTe₂ on CaF₂ as compared to the estimation using RHEED pattern, and to understand the preference of in-plane rotational orientation, grazing incidence X-ray diffraction (GI-XRD) was done. GI-XRD from sample A shows (see Fig. 4(a)) an extended line corresponding to overlapped $\{10\overline{1}0\}$ and $\{10\overline{1}1\}$ set of planes of MoTe₂ and the sharp high intensity peak is from the CaF₂ substrate. The in-plane lattice constant of 2H-MoTe₂ calculated from the $\{10\overline{1}0\}$ peak corresponds to 3.638 Å. The full width half maximum (FWHM) of 2H-MoTe₂ { $10\overline{1}0$ } peak is calculated to be ~0.079 Å⁻¹ and the direct beam FWHM is ~0.009 Å⁻¹. By subtracting the width of the direct beam, the genuine FWHM of the $\{10\overline{1}0\}\$ peak was estimated to be ~0.07 Å⁻¹ which translates to a grain size of \sim 92 Å [30]. From Fig. 4(b), which is an inplane phi (ϕ) scan, we can see that MBE MoTe₂ undergoes significant twinning thus showing 2 sets of 6-fold symmetry diffraction patterns. The peak ratio between adjacent peaks separated by 30° is ~ 0.5 . This shows that almost 30% of the grains are twins. Also the wide FWHM (6.7–7.2°) of these peaks signifies a large deviation of grains from the preferred orientation. From measurements and simulation it has been shown in 2*H*-MoSe₂ that $\{10\overline{1}1\}$ and $\{10\overline{1}2\}$ peaks are ~ 10 times weaker than $\{10\overline{1}3\}$ [31]. 2*H*-MoTe₂ having the same crystal structure as 2H-MoSe₂, we also observe the $\{10\overline{1}3\}$ set of peaks at higher q₁ (see SI Fig. 5(b)), from which the out of plane lattice constant (c-spacing) is calculated to be 14.4 Å. The c axis lattice constant obtained from cs-TEM as shown



Fig. 4. For sample A (a) in-plane ω - 2θ scan showing a peak corresponding to CaF₂ and {1010} of MoTe₂. The extended streak in the perpendicular direction is due to overlap of the extended rods from {1010} and {1011} in reciprocal space due to the ~5 monolayer thin film. Below it is the integrated intensity in a range of 0.02–0.15 Å⁻¹ q_⊥ corresponding to {1010} peak (b) The in-plane phi(ϕ) scan of the {1010} peak of MoTe₂ to understand the rotational alignment and the extent of twinning in the grown film. Below it is the integrated intensity in a range of 0.02–0.15 Å⁻¹ q_⊥.

in Fig. 5(b) is 13.9 Å. The reported value for the inplane lattice constant and c axis lattice constant from bulk 2H-MoTe₂ are 3.52 Å and 13.966 Å, respectively [10]. To understand this discrepancy, we compare the intensity along q_{\perp} (see SI Fig. 7with simulations and observation for various polytypes of NbSe₂ by Toshihiro Shimada et al. [32] 2H-MoTe₂ and 2H-NbSe₂ share identical in-plane crystal structures and hence, would give similar intensity profiles along q_{\perp} for various stacking orders (polytypes). Our q_{\perp} (see SI Fig. 7scan closely matches the one reported by Toshihiro Shimada et al. [32] on Se-GaAs, which is explained as a combination of 2Hb and 3R NbSe₂. 2Hb and 3R both have the trigonal prismatic monolayer but the stacking sequence is different, with 3R having a 3-layer periodicity as compared to 2-layer for 2Hb. The value for MBE 2H-MoTe₂ obtained by TEM is spatially local but X-ray beam for GI-XRD has a foot print of $\sim 2 \text{ mm} \times 10 \text{ mm}$.

Both, the excess Te (>2 Te:Mo film stoichiometry) measured using XPS and the presence of stacking faults resulting in mixture of 2*Hb* and 3*R* phases could be the contributing factors for the observed larger 'a' and 'c' lattice constants in MBE 2*H*-MoTe₂ compared to bulk 2*H*-MoTe₂.

SI Fig. 6(b) shows the cs-TEM of the mixed phase sample B. Comparing the film thickness for sample A and sample B (see SI Fig. 6(a and b)), they are very similar but apart from lattice spacing, phase composition from sample B is not discernible. 1T'- MoTe₂ is known to crystallize in $P2_1/m$ space group with lattice constants of a = 6.33 Å, b = 3.48 Å and c = 13.82 Å [10], where b is very close to the lattice constant of 2H-MoTe₂. Therefore, for the sample C since the RHEED shows polycrystalline rings, irrespective of the in-plane rotational orientation of the film, if it was purely 1T', we would expect to observe a ring corresponding to the {010} set of planes in a similar scan as for Sample A (Fig. 4(a) or SI Fig. 5(a)). SI Fig. 5(c)shows the GI-XRD on sample C ie. UP-MoTe_x. SI Fig. 5(c) shows that there is no signal observed corresponding to 1*T*-MoTe₂, the only peak is corresponding to CaF₂. This is likely due to extremely low signal from the polycrystalline highly defective 1T-MoTe₂ thin film or the film has a new unknown phase with a different crystal structure. Finally, the GI-XRD exercise also shows the variability in crystallinity of CaF₂ from substrate to substrate, motivating the use of epi-ready substrates such as, GaAs.

3.4.2. MoTe₂ films on GaAs substrate

It is very interesting to note that after cooling to room temperature, although the films on CaF₂ (Fig. 5(a)) only show \sim 10 nm droplet like structures that are \sim 4–5 nm high, sample D shows 100s nm long Te crystallites on the surface. The height of these crystallites is \sim 9 nm and they are about \sim 50 nm wide. From AFM image



Fig. 5. Sample A: (a) The surface of the thin film post-growth on sample A and (b) cs-TEM of 2H-MoTe₂ on CaF₂. Sample D: (c) The surface of the thin film of 2H-MoTe₂ on GaAs post-growth shows several tellurium crystallites, surprisingly with preferential direction of orientation. (d) cross-sectional TEM shows the abrupt interface between GaAs and MoTe₂, better quality of ~9 monolayer MoTe₂ than on CaF₂ and pure tellurium crystallite with a significantly different lattice constant and contrast. The pair of white lines and the pair of red lines are a guide to the eye marking the difference in lattice constant of Te and 2H-MoTe₂.

(Fig. 5(c)) and SEM image (SI Fig. 8(a)) we can observe that these crystallites have preferential crystallographic orientation with the underlying GaAs (111) with triangular symmetry. The fact that these crystallites are purely tellurium is confirmed by the markedly distinct lattice spacing compared to MoTe₂ as seen in the high resolution transmission electron microscopy (HRTEM) image shown in Fig. 5(d). This is further confirmed by TEM Energydispersive X-ray spectroscopy (not shown). The HRTEM image (Fig. 5(d)) also shows a high quality of MoTe₂ with a c-axis lattice spacing of 13.9 Å consistent with 2H-MoTe₂. These crystallites could have likely been formed during the cooling process to room temperature. We don't observe such Te crystallite formation on films on CaF₂ (see SI Fig. 3(a)). XRD scan in SI Fig. 8(b)shows that at room temperature the (004) peak for sample D appears at 24.46°, which is lower than that for bulk 2H-MoTe₂ and 1T-MoTe₂ [10]. It corresponds to a c-spacing of 14.52 ± 0.05 Å. It is analyzed below along with (004) peak of sample E (see Table 4).

From the phase diagram [9], one might expect mixed phase formation during the post-growth anneal at 550 °C under Te. LEED and XRD was done to check the phase of the film grown. Prior to LEED, the Se cap was desorbed by annealing the sample in UHV at 300 °C for \sim 30 min. A LEED pattern measured with 40 eV electrons (Fig. 6(a) shows two sets of spots with the outer hexagonal pattern corresponding to the lattice constant of 2H-MoTe₂ ('a' lattice constant using LEED = 3.57 ± 0.03 Å) and the inner pattern corresponding to an effective 2×2 superstructure. This is very interesting because it is consistent with the second set of RHEED streaks seen for sample D in Fig. 1(e-f), discussed previously. One possible explanation for the extra spots is a change in the surface periodicity relative to the bulk 2H-MoTe₂ crystal structure due to a reconstruction or the presence of ordered defects. An alternate possibility is that the large electron spot size ($\sim 1 \text{ mm}$) may be averaging over three domains of 1T'-MoTe₂ rotated by 60 degrees and 120 degrees (see SI Fig. 9).

Table 4	
In-plane and out of plane lattice constants for the 2H-MoTe ₂ samples in this study obtained by various to	echniques.

Sample ID	RHEED (Å)	XRD (Å)	GI-XRD (Å)	LEED (Å)	TEM (Å)
А	$a = 3.5 \pm 0.1$		$a \texttt{=} 3.64 \pm 0.03$		c = 13.9 \pm 0.1
			c = 14.4 ± 0.03		
D	a = 3.6 \pm 0.05	c = 14.52 \pm 0.05			c = 13.8 \pm 0.1
Е	a = 3.5 ± 0.1	c = 14.25 ± 0.04		a = 3.57 ± 0.03	
Bulk 2H		a = 3.52085 [10]			
		c = 13.9664 [10]			
Bulk 1T'		a = 6.3274 [10]			
		b = 3.4755 [10]			
		c = 13.8100 [10]			



Fig. 6. (a) LEED from sample E post-decapping without any air exposure, the red spots are guide to the eye demonstrating the 2×2 superstructure (b) Temperature dependent XRD on sample E under nitrogen environment showing the phase of MoTe₂ and its thermal stability.

To distinguish between these possibilities, temperature dependent XRD was done on sample E. At room temperature (002), (004), (006) and (008) peak 2θ positions of the as grown film are 12.17°, 25.02°, 38.02° and 51.38° respectively. The reported room temperature 2θ positions for the (002), (004), (006) and (008) peaks for 2H-MoTe₂ are 12.66°, 25.48°, 38.63° and 52.34° respectively. Keum et al. [10] report using temperature dependent XRD that the (004) 2θ peak of 2H-MoTe₂ is at 25.5° at room temperature and it shifts to slightly greater than 26° at temperatures above 600 °C corresponding to (004) 2θ peak of 1T'-MoTe₂. In our case, the $(004) 2\theta$ peak position is much lower than both peaks $(\sim 0.5^{\circ}$ lower than 2*H*-MoTe₂), which corresponds to a c-spacing of 14.25 ± 0.04 Å, and the peaks from the film is lost above 400 °C. Therefore, without a chalcogen over pressure, MBE grown fewlayer 2H-MoTe₂ dissociates between 400 °C and 500 °C (see Fig. 6 (b)). Also, sample D (see SI Fig. 8(b)) which was grown without anneal at 550 $^\circ\text{C}$ anneal has ${\sim}0.5^\circ$ lower than sample E. Peak at smaller 2θ implies larger lattice constant but its origin is unclear yet. One possible explanation is the presence of excess tellurium in the crystal, which has been previously reported for bulk crystals [31]. The MoTe₂ phase diagram [9] shows that 2H-MoTe₂ is not a line compound. The 2H phase of MoTe₂ can be formed in spite of a 1% sub- or super-stoichiometric incorporation of tellurium. Hence, as XRD for both samples on GaAs does not show detectable peaks from the 1T-MoTe₂ phase, the 2×2 superstructure observed in LEED, is likely a surface feature rather than the presence of rotated domains of 1T'-MoTe₂. Lattice constants obtained from the various techniques described above have been tabulated in Table 4.

3.5. Electrical measurements

Sheet resistivity of the sample A and sample C were measured to be 5468 Ω/\Box and 13,255 Ω/\Box , respectively. This is of interest because film on sample A is 2*H*-MoTe₂ and film on sample C is assigned to *UP*-MoTe_x. 1*T*'-MoTe₂ is metallic and is expected to have lower resistance than 2*H*-MoTe₂. From XPS of sample C, we observe that more Mo is oxidized than that is bound to Te, so, weather *UP* phase is highly defective 1*T*'-MoTe₂ or a new unknown phase, this extensive oxidation could be the cause of the significant increase in resistivity. TeO₂ glasses show semiconducting behavior [33].

In order to test the electrical characteristic of the transferred MBE grown MoTe₂ film on GaAs, contacts were made to the transferred flake on to SiO₂ (SI Fig. 10(a)). Using backgating, no modulation was observed (see SI Fig. 10(b and c)). Excess Te in the film could a cause for this behavior. But even if techniques are devised to overcome excess Te incorporation, its important to note that using XPS, oxidation of 8% of the surface area in 20 min was observed and, through the process of exfoliation and fabrication the film is likely to undergo extensive oxidation if appropriate capping is not done.

4. Conclusion

This work employs extensive large area structural and chemical characterizations of MBE grown few layer MoTe₂, with complementing electrical characterization. We show that for growth of few layer 2H-MoTe₂ at a low temperature of 340 °C and growth rate of ~6 min/ML, we need an incident Te:Mo flux greater than 100. The 2*H* phase growth on CaF₂ and GaAs is confirmed using RHEED, Raman and XPS, but the Te:Mo stoichiometry determined by XPS. GI-XRD shows a small grain size of \sim 90 Å, twinning and a higher-than-expected 'a' and 'c' spacing for MBE 2H-MoTe₂ on CaF₂. XRD on MBE 2H-MoTe₂ on GaAs also shows larger c spacing than both bulk 2H-MoTe₂ and bulk 1T'-MoTe₂. On CaF₂, greater than 2 Te: Mo stoichiometry determined by XPS and on GaAs, Te crystallite formation on the surface, and a 2×2 pattern in RHEED and LEED have been observed. All these have been hypothesized as signs of excess Te incorporation into few-layer 2H-MoTe₂ during growth using elemental Mo and uncracked Te sources. At ambient pressure in N_2 atmosphere, MBE 2*H*-MoTe₂ on GaAs is only stable up to 300 °C. Excess Te in the film can explain the high electrical conductivity, non-modulating behavior and easy dissociation of the film with increasing temperature prior to phase transition to the 1T^r phase, a more stable phase at higher temperatures. Finally, we demonstrate the swift oxidation ($\sim 8\%$ surface area in 20 min) of the MBE MoTe₂ film on exposure to air. With the various complementing large area and local characterizations, this study has provided insight into the few layer MBE growth of Mo-Te system on 3D substrates. We believe our work motivates study into new phases obtained under UHV conditions as well as into techniques to directly probe and overcome excess Te incorporation.

5. Experimental methods

5.0.1. Raman spectroscopy

Raman measurements were performed in the backscattering configuration using a WITec Alpha 300 system at room temperature. Measurement was done using a $100 \times$ objective, 1800 grooves/mm grating, 488 nm laser and 0.75 mW power.

5.0.2. X-ray photoelectron spectroscopy

XPS on the CaF₂ samples was carried out ex-situ using a monochromated Al K α source (h ν = 1486.7 eV) and an Omicron Argus detector (MCD-128) operating with a pass energy of 15 eV. XPS spectra were acquired at a pass energy of 15 eV and take-off angle (defined with respect to the sample surface) of 45° and 80°. For XPS peak deconvolution, the spectral analysis software Aanalyzer was employed, where Voigt line shapes and an active Shirley background were used for peak fitting [34].

XPS on the 2*H*-MoTe₂/n⁺GaAs samples was measured using a non-monochromated Al K α source and a Scienta R4000 electron analyzer operating at a pass energy of 100 eV. All spectra were measured at normal emission, i.e. 90 degrees relative to the sample surface. Central peak locations were determined by Lorentzian fits with linear backgrounds.

5.0.3. X-ray diffraction

Out of plane XRD and temperature dependent XRD on the 2H-MoTe₂/GaAs samples is done using the Rigaku SmartLab X-ray Diffractometer with Cu K α X-ray source. The GI-XRD is done using the G2 hutch at the CHESS beamline (http://www.chess.cornell. edu/gline/G2.htm), operating with a X-ray energy of 11.31 keV.

5.0.4. Transmission electron microscopy

The atomic structure analysis for sample A and sample D was carried out on FEI Titan 80–300 Transmission Electron Microscope operated at 300 kV.

TEM on sample B was done using JOEL ARM200F atomic resolution analytical microscope.

5.0.5. LEED

LEED on the 2*H*-MoTe₂/n⁺GaAs samples was measured using a Specs ErLEED 3000 system with an incident electron energy of 40 eV. The electron spot size was approximately 1 mm in diameter, and the total angular field of view was 100 degrees. Following the Se decapping at 300 °C, in situ LEED and XPS measurements were both performed at room temperature in an analysis chamber with pressure below $\sim 1 \times 10^{-10}$ Torr.

Acknowledgments

This work was supported in part by the Center for Low Energy Systems Technology (LEAST), one of six centers of STARnet, a Semiconductor Research Corporation program sponsored by MARCO and DARPA. This work made use of the Cornell Center for Materials Research Shared Facilities which are supported through the NSF MRSEC program (DMR-1120296). This work is based upon research conducted at the Cornell High Energy Synchrotron Source (CHESS) which is supported by the National Science Foundation and the National Institutes of Health/National Institute of General Medical Sciences under NSF award DMR-1332208. MBE growth is in part supported by NSF Grant DMR 1400432 and NSF-EFRI 2DARE Grant DMR 1433490.

Appendix A. Supplementary material

Supplementary data associated with this article can be found, in the online version, at https://doi.org/10.1016/j.jcrysgro.2017.10. 024.

References

- [1] K. Deng et al., Nat. Phys. 12 (2016) 1105–1111.
- [2] C. Gong, H. Zhang, W. Wang, L. Colombo, R.M. Wallace, K. Cho, Appl. Phys. Lett. 103 (2013) 053513.
- [3] H.C. Diaz, Y. Ma, R. Chaghi, M. Batzill, Appl. Phys. Lett. 108 (2016) 191606.
- [4] J. Chen, G. Wang, Y. Tang, J. Xu, X. Dai, J. Jia, W. Ho, M. Xie, 2016. Available from: arXiv preprint <arXiv:1612.06105>.
- [5] R. Schlaf, O. Lang, C. Pettenkofer, W. Jaegermann, J. Appl. Phys. 85 (1999) 2732– 2753.
- [6] R. Yan, S. Fathipour, Y. Han, B. Song, S. Xiao, M. Li, N. Ma, V. Protasenko, D.A. Muller, D. Jena, H.G. Xing, Nano Lett. 15 (2015) 5791–5798.
- [7] M.O. Li, D. Esseni, J.J. Nahas, D. Jena, H.G. Xing, IEEE J. Electron Dev. Soc. 3 (2015) 200–207.
- [8] M.O. Li, D. Esseni, D. Jena, H.G. Xing, Lateral transport in two-dimensional heterojunction interlayer tunneling field effect transistor (Thin-TFET), in: Device Research Conference (DRC), 2014 72nd Annual, 2014, pp. 17–18.
- [9] L. Brewer, R. Lamoreaux, Binary Alloy Phase Diagrams, second ed., Ed. T.B. Massalski, ASM International, Materials Park, Ohio, vol. 3, 1990, pp. 2675– 2676.
- [10] D.H. Keum, S. Cho, J.H. Kim, D.-H. Choe, H.-J. Sung, M. Kan, H. Kang, J.-Y. Hwang, S.W. Kim, H. Yang, K.J. Chang, Y.H. Lee, Nat. Phys. 11 (2015) 482–486.
- [11] Y. Yu, G. Wang, S. Qin, N. Wu, Z. Wang, K. He, X.-A. Zhang, Carbon 115 (2017) 526-531.
- [12] S. Cho, S. Kim, J.H. Kim, J. Zhao, J. Seok, D.H. Keum, J. Baik, D.-H. Choe, K. Chang, K. Suenaga, S.W. Kim, Y.H. Lee, H. Yang, Science 349 (2015) 625–628.
- [13] J. Bernede, C. Amory, L. Assmann, M. Spiesser, Appl. Surface Sci. 219 (2003) 238–248.
- [14] W. Jaegermann, D. Schmeisser, Surface Sci. 165 (1986) 143-160.
- [15] F. Gamble, J. Solid State Chem. 9 (1974) 358–367.
- [16] L. Zhou, K. Xu, A. Zubair, A.D. Liao, W. Fang, F. Ouyang, Y.-H. Lee, K. Ueno, R. Saito, T. Palacios, J. Kong, M.S. Dresselhaus, J. Am. Chem. Soc. 137 (2015) 11892–11895.
- [17] J.C. Park, S.J. Yun, H. Kim, J.-H. Park, S.H. Chae, S.-J. An, J.-G. Kim, S.M. Kim, K.K. Kim, Y.H. Lee, ACS Nano 9 (2015) 6548–6554.
- [18] J. Zhou et al., Adv. Mater. 1603471 (1-8) (2017).
- [19] J. Bernede, M. Kettaf, A. Khelil, M. Spiesser, Phys. Status Solidi (A) 157 (1996) 205–209.
- [20] H. Zhu, Q. Wang, C. Zhang, R. Addou, K. Cho, R.M. Wallace, M.J. Kim, Adv. Mater. (2017).
- [21] S. Vishwanath, X. Liu, S. Rouvimov, L. Basile, N. Lu, A. Azcatl, K. Magno, R.M. Wallace, M. Kim, J.-C. Idrobo, J.K. Furdyna, D. Jena, H.G. Xing, J. Mater. Res. 31 (2016) 900–910.
- [22] A. Roy, H.C. Movva, B. Satpati, K. Kim, R. Dey, A. Rai, T. Pramanik, S. Guchhait, E. Tutuc, S.K. Banerjee, ACS Appl. Mater. Interf. 8 (2016) 7396–7402.
- [23] S. Vishwanath, X. Liu, S. Rouvimov, P.C. Mende, A. Azcatl, S. McDonnell, R.M. Wallace, R.M. Feenstra, J.K. Furdyna, D. Jena, H.G. Xing, 2D Materials 2 (2015) 024007.
- [24] A. Koma, K. Saiki, Y. Sato, Appl. Surface Sci. 41 (1990) 451-456.
- [25] C. Adenis, V. Langer, O. Lindqvist, Acta Crystallogr. Sect. C: Cryst. Struct. Commun. 45 (1989) 941–942.
- [26] A. Pine, G. Dresselhaus, Phys. Rev. B 4 (1971) 356.
- [27] C.H. Naylor, W.M. Parkin, J. Ping, Z. Gao, Y.R. Zhou, Y. Kim, F. Streller, R.W. Carpick, A.M. Rappe, M. Drndic, J.M. Kikkawa, A.T.C. Johnson, Nano Lett. 16 (2016) 4297–4304.
- [28] F. Werfel, E. Minni, J. Phys. C: Solid State Phys. 16 (1983) 6091.
- [29] J.H. Park, S. Vishwanath, X. Liu, H. Zhou, S.M. Eichfeld, S.K. Fullerton-Shirey, J.A. Robinson, R.M. Feenstra, J. Furdyna, D. Jena, H.G. Xing, ACS Nano 10 (2016) 4258–4267.
- [30] T. Shimada, H. Nishikawa, A. Koma, Y. Furukawa, E. Arakawa, K. Takeshita, T.-I. Matsushita, Surface Sci. 369 (1996) 379–384.
- [31] P. James, M. Lavik, Acta Crystallogr. 16 (1963), 1183-1183.
- [32] T. Shimada, Y. Furukawa, E. Arakawa, K. Takeshita, T. Matsushita, H. Yamamoto, A. Koma, Solid State Commun. 89 (1994) 583–586.
- [33] R. Hampton, W. Hong, G. Saunders, R. El-Mallawany, J. Non-crystalline Solids 94 (1987) 307–314.
- [34] A. Herrera-Gomez, A. Hegedus, P. Meissner, Appl. Phys. Lett. 81 (2002) 1014– 1016.